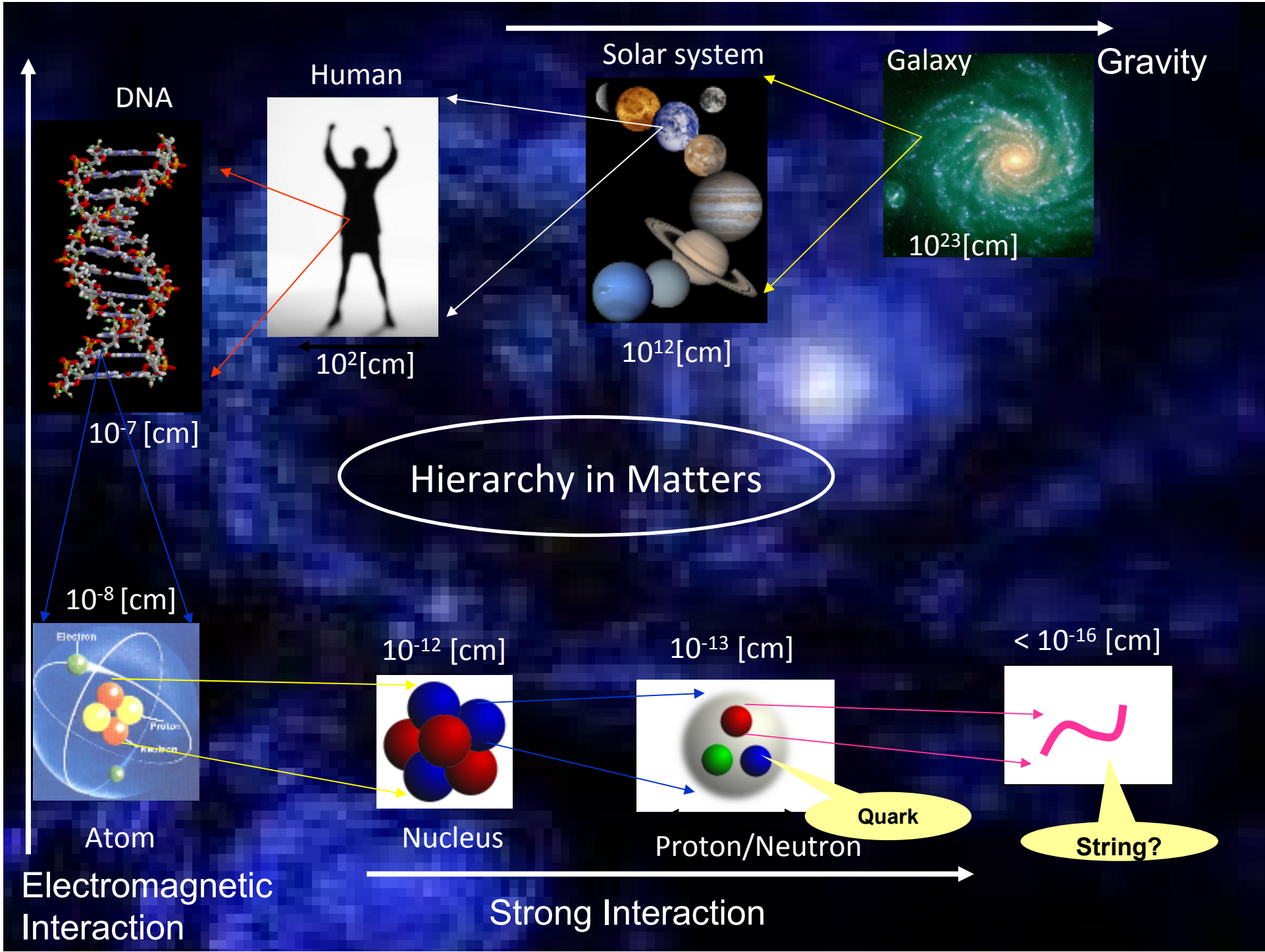


Nuclear Theory

原子核理論





Discovery of radioactivity (End of 19th century)

- 1895 Discovery of X-ray (Roentgen)
- 1896 Natural radioactivity (Becquerel)
alpha-ray from Uranium
- 1897 Discovery of electron (Thomson)
- 1898 Polonium, Radium (Currie)
- 1900 alpha, beta, gamma (Rutherford)

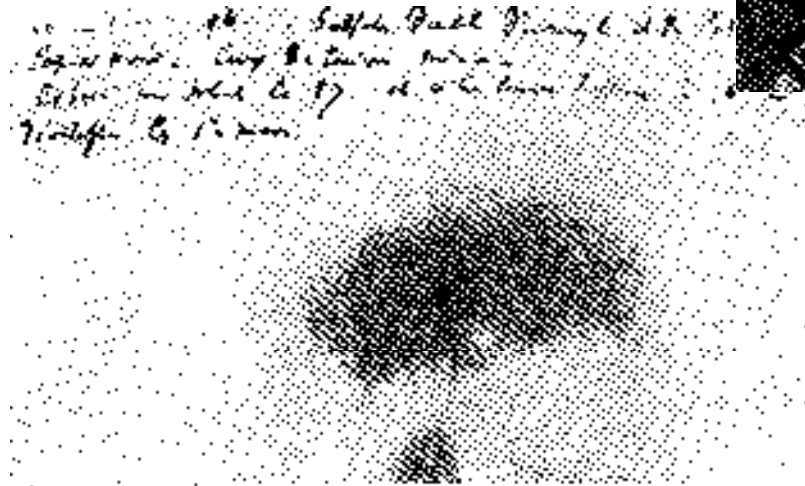
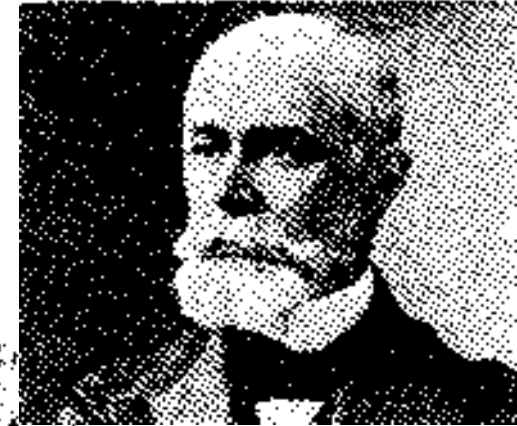


図 3.1 天然放射線の発見の最初の瞬間。これは1896年2月26日に撮影された。それをベッケルの10月1日に撮影した写真と比較して、その放射線は、これは明らかで、ウランの原子は連続的に放射線を出していることが観察された。この発見は *Comptes rendus de l'Académie des Sciences de Paris* [122, 301 (1896)] に載った。(CNA)

- 1911 Discovery of nucleus (Rutherford)
- 1932 Discovery of neutron (Chadwick)

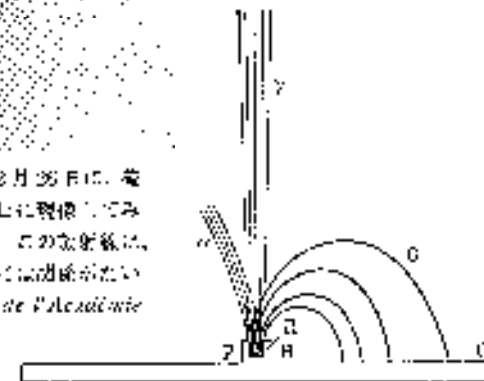
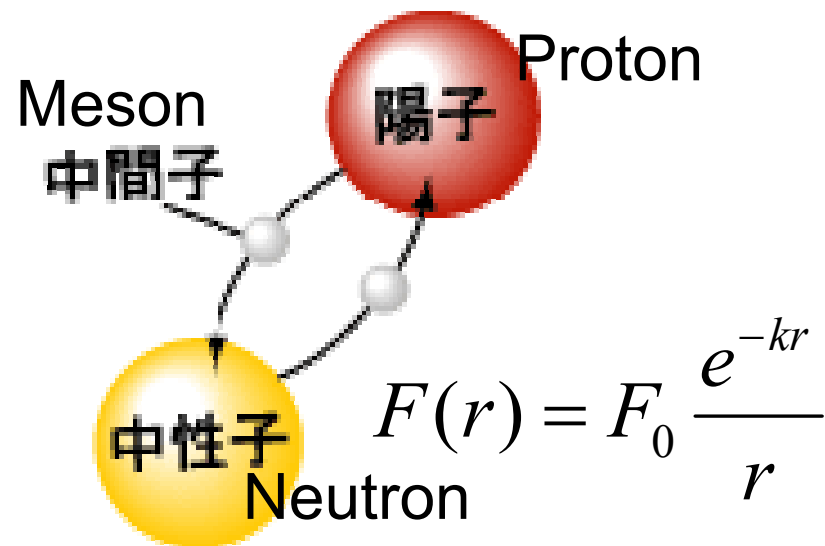
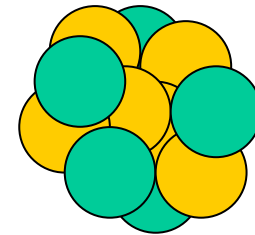


図 3.2 三種の放射線の透過。この図は、放射線が磁場を通過する様子を示している。放射線は、磁場の作用によって異なる方向に曲げられる。α線は最も大きく曲げられ、β線は最も小さく曲げられる。γ線は曲げられない。(CNA)

Nuclear interaction: Force to bind nucleus

Protons and neutrons bound together to form a nucleus.



1934: Meson theory
(Hideki Yukawa)



中間子をやりとりすることで
「陽子と中性子の間には力がはたらき近づいている」

Quarks and gluons are described by the QCD.

How small is the nucleus?

Estimate the Compton wave length of pion

$$\lambda_{\pi} = \frac{\hbar}{m_{\pi}c} = ?$$

This is a typical magnitude of range of nuclear force

$$F(r) \sim F_0 \frac{\exp(-r / \lambda_{\pi})}{r}$$

and the typical size of nucleus.

Nuclear Physics ~ Femto Physics

How strong is the nuclear force?

There is a deuteron (p+n) that is a bound state of a proton and a neutron.

The range of the force is roughly the pion Compton wave length.

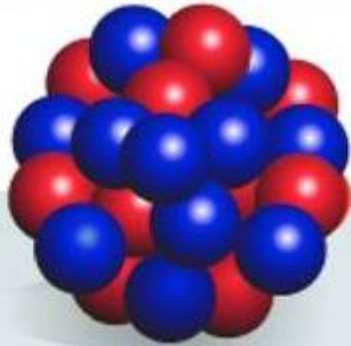
(1) Estimate order of magnitude using the uncertainty principle

$$\Delta x \cdot \Delta p \sim \hbar$$

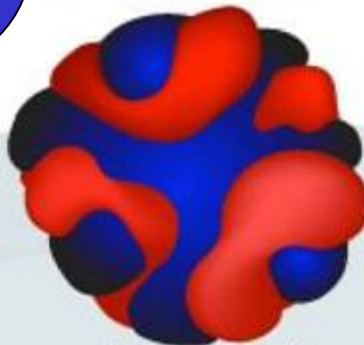
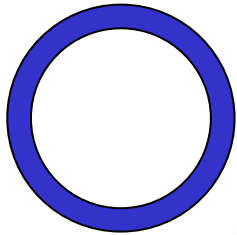
$$Mc^2 \sim 1 \text{ GeV} = 1000 \text{ MeV}$$

(2) Estimate order of nuclear time scale

Image of nucleus



protons, neutrons



nucleonic densities
and currents



Liquid-drop Model

Binding energy

$$B/A \approx 8 \text{ MeV}$$

Density

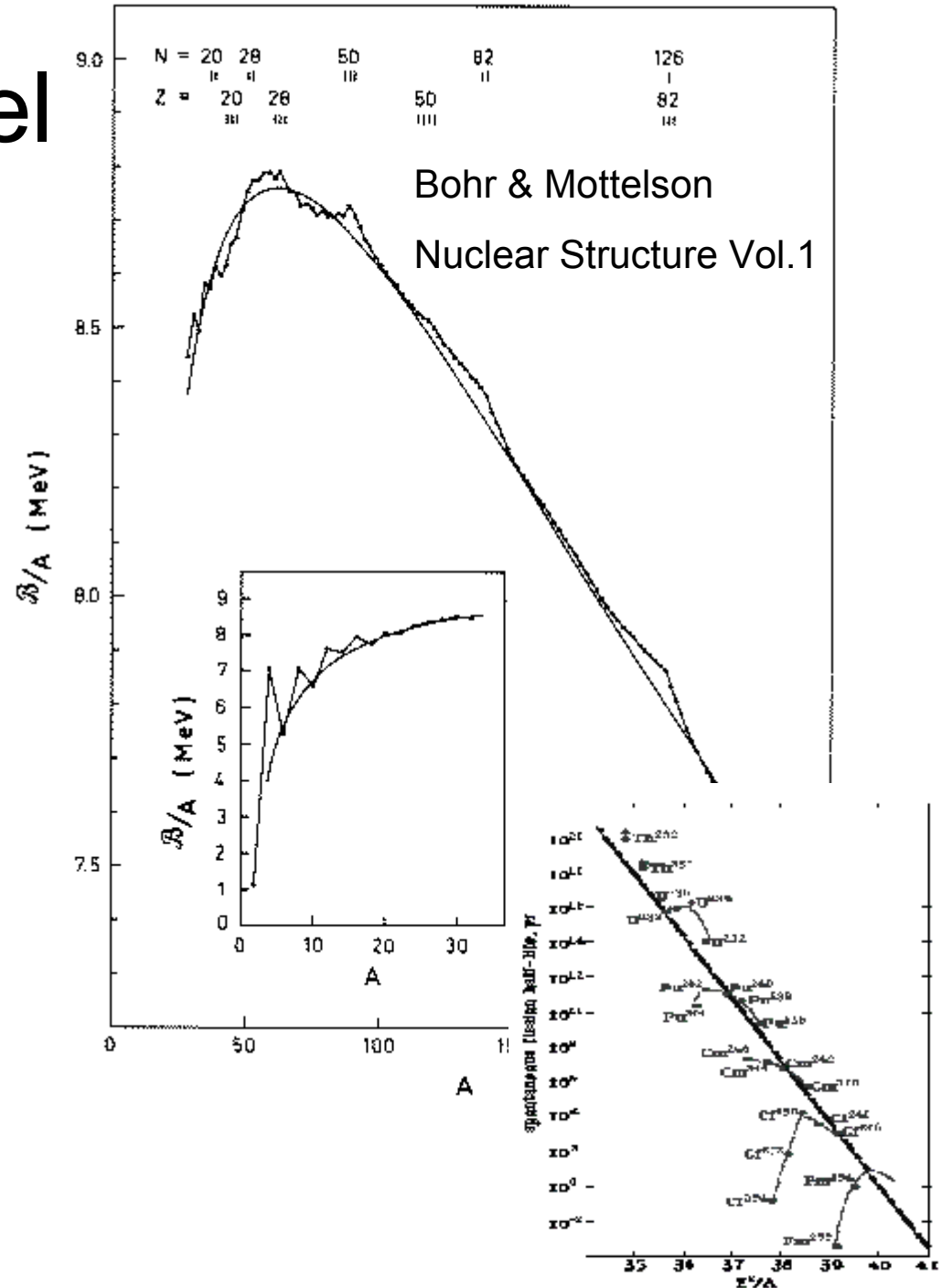
$$\rho \approx 0.14 \text{ fm}^{-3} \quad d \approx 2 \text{ fm}$$

Bethe-Weizsäcker mass formula

$$B(N, Z) = a_V A - a_S A^{2/3} - a_{sym} \frac{(N - Z)^2}{A} - a_C \frac{Z^2}{A^{1/3}} + \delta(A)$$

Nuclear fission

$$x = \frac{E_C}{2E_S} \sim \frac{Z^2}{A}$$



Nucleus as a quantum liquid

- Classical vs Quantum
 - Strength of interaction vs Zero-point kinetic energy

$$V_0 \quad \text{vs} \quad \frac{\hbar^2}{2Mc^2}$$

c : Length scale of the interaction

V_0 : Energy scale of the interaction

Nuclear force vs molecular force

Bohr, Mottelson, Nucl. Str. Vol.1

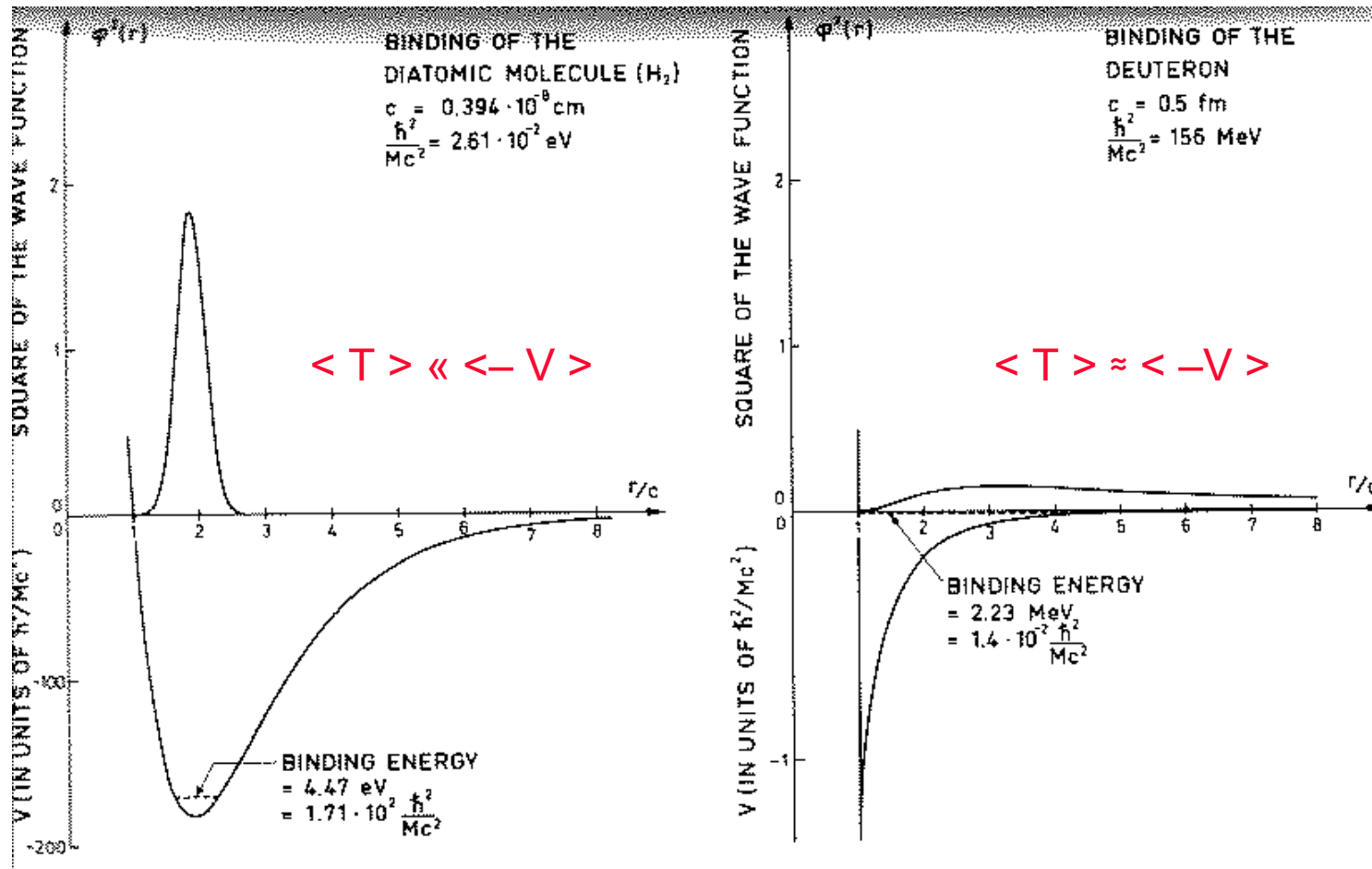


Figure 2-36 The molecular interaction corresponds to a "Morse potential" $V(r) = D[1 - \exp(-a(r - r_0))]^2$ with the constants adjusted.

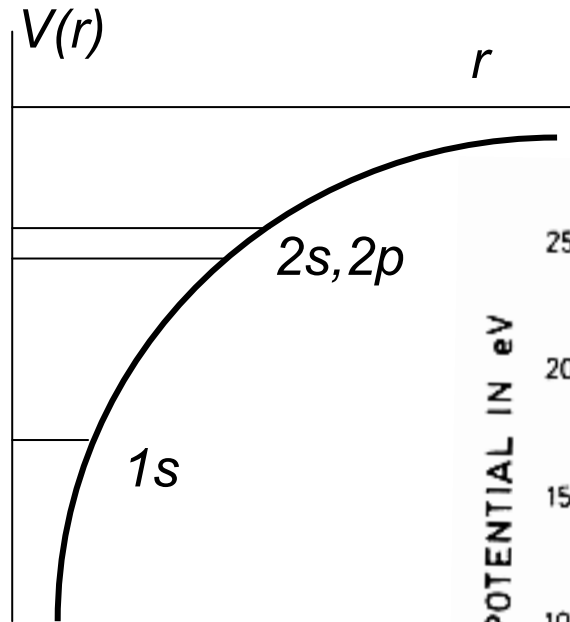
Crystallized at low temperature

Classical \rightarrow MD

Liquid at low temperature

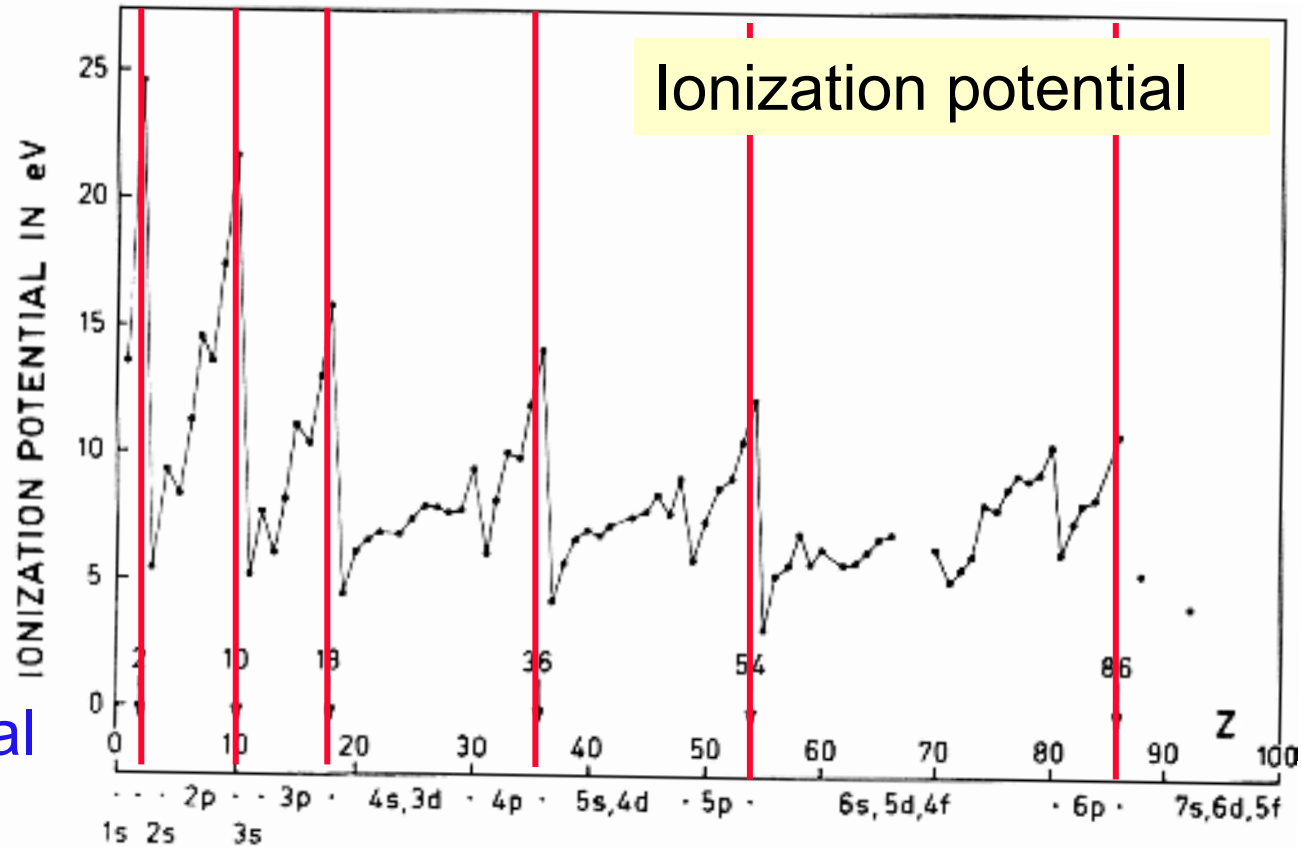
Quantum

Electronic single-particle motion in atoms



Single-particle orbitals in the Coulomb potential → **Magic number**

Free particles in Coulomb potential

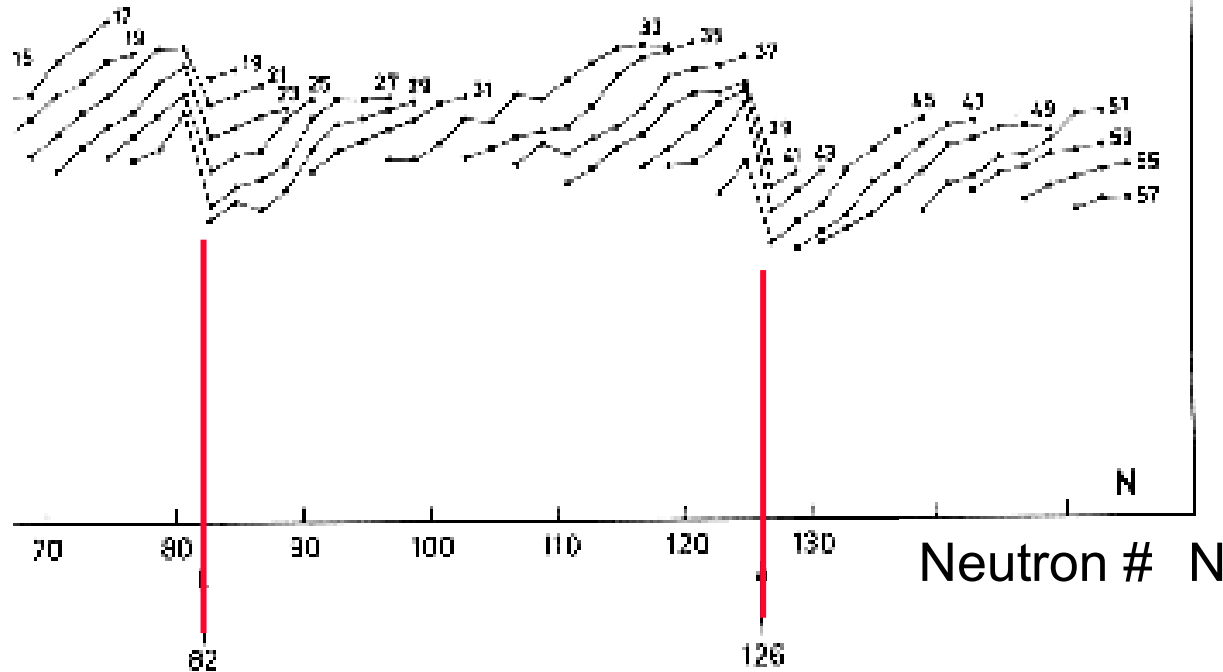


Nucleonic single-particle motion in nucleus

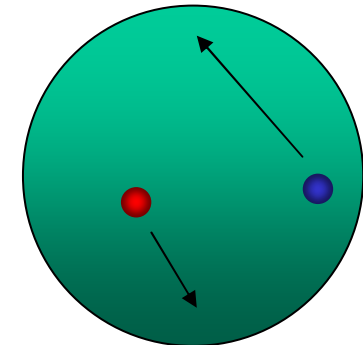
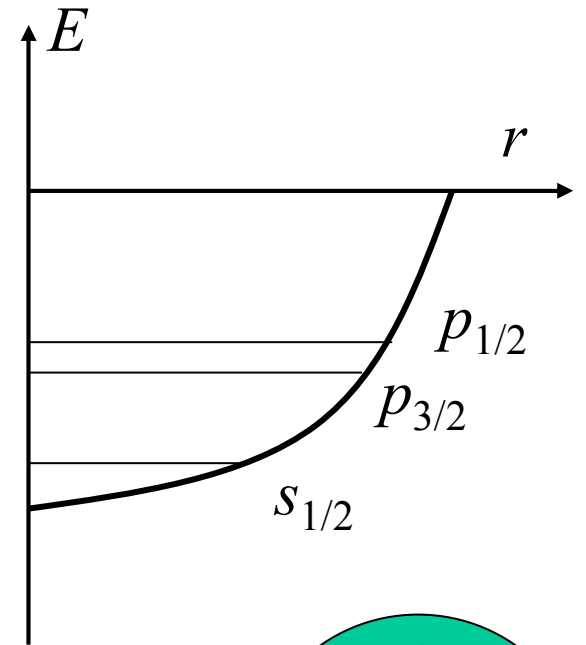
Bohr & Mottelson, Nuclear Structure Vol.1

Neutron Separation energy

$$S_n(N, Z) = \beta(N, Z) - \beta(N-1, Z) \quad \begin{array}{l} N \text{ odd} \\ Z \text{ even} \end{array}$$



Shell Model (Mayer-Jensen)

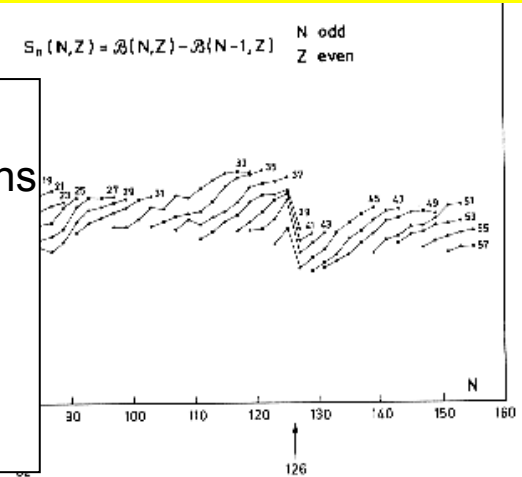
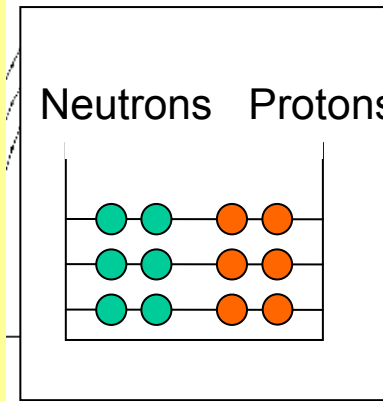
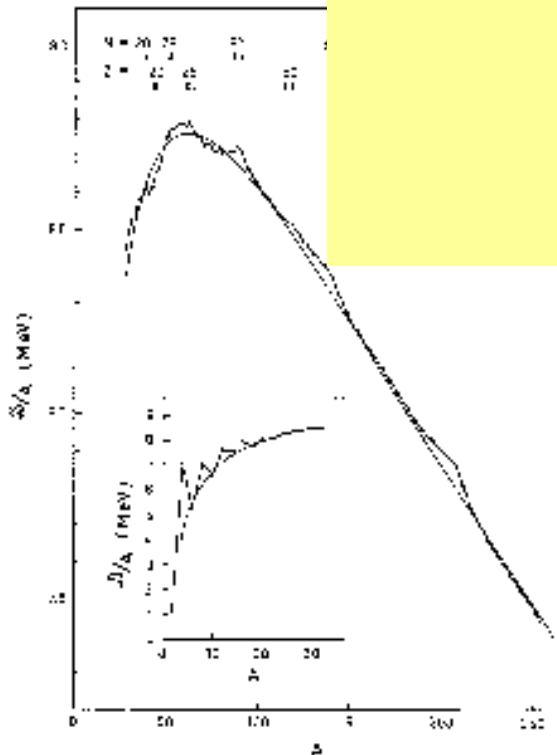


Nucleus is liquid or gas?

Liquid drop model
(Bethe-Weizsäcker)

Shell model
(Mayer-Jensen)

$$B(N, Z) = a_V A - a_S A^{2/3} - a_{sym} \frac{(N - Z)^2}{A} - a_C \frac{Z^2}{A^{1/3}} + \delta(A)$$



$$V(r) = \frac{1}{2} M \omega^2 r^2 + v_{ll} \ell^2 + v_{ls} \vec{\ell} \cdot \vec{s}$$

→ Symmetry breaking in the Unified Model
(Bohr-Mottelson)

Different reaction rate

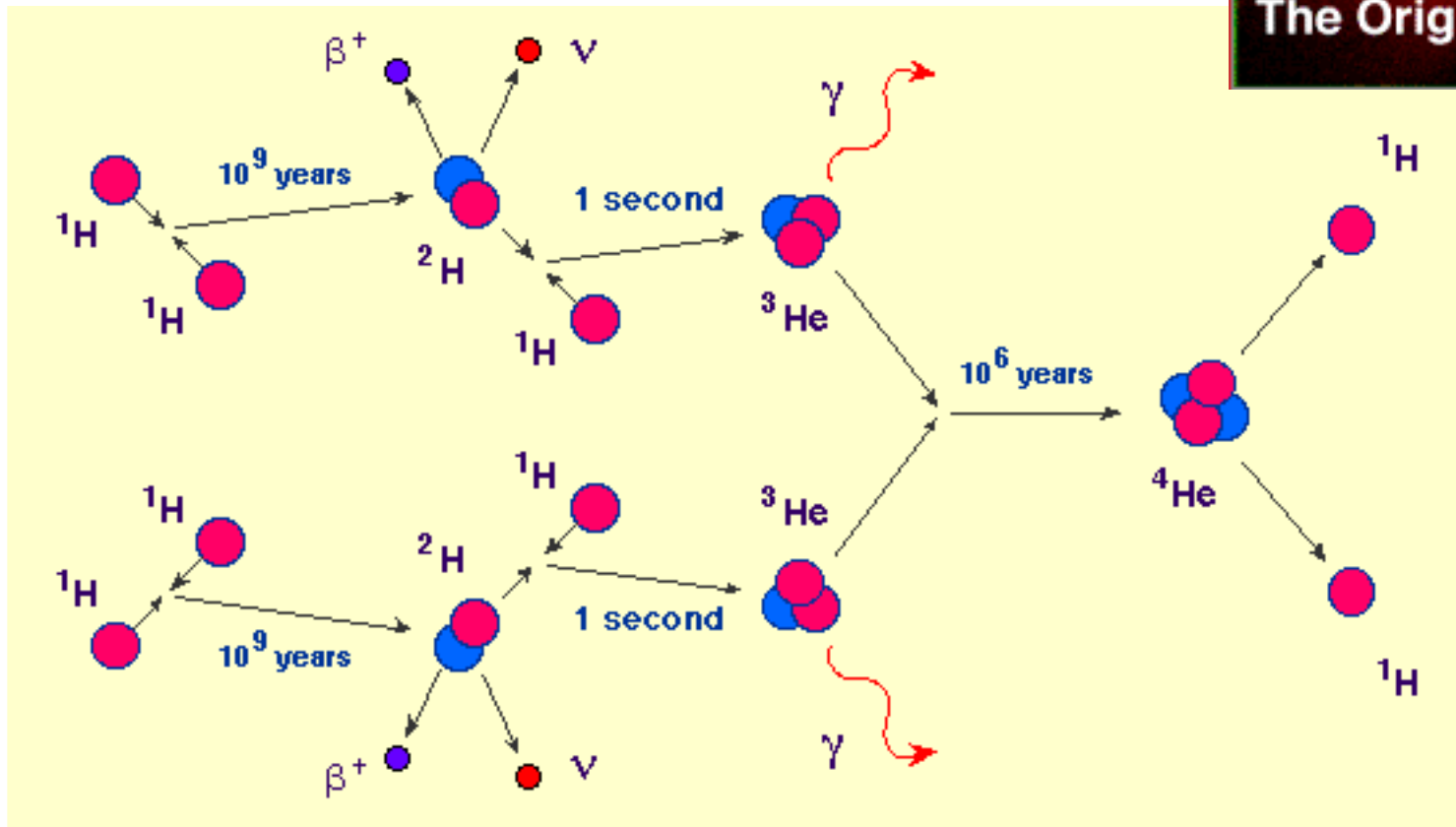
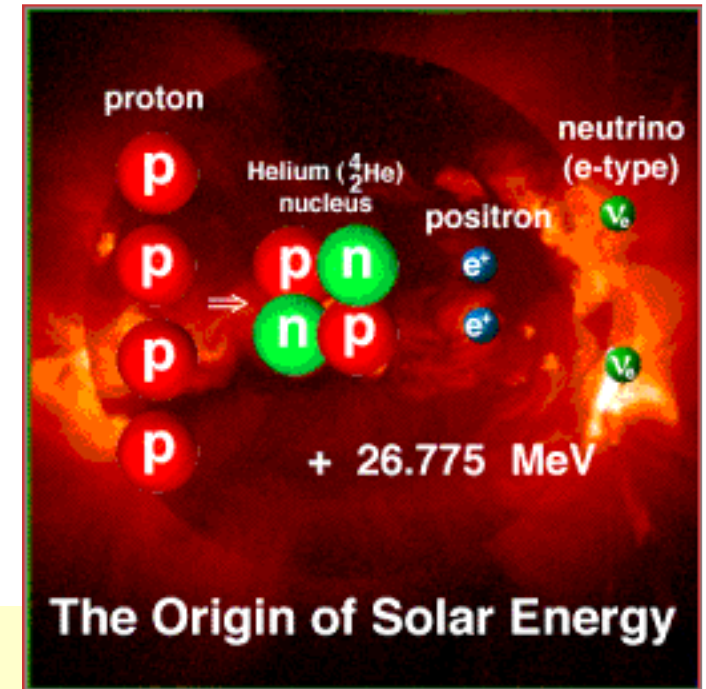
- Transfer reaction (Strong interaction)
 - $^{15}\text{N}(p, \alpha)^{12}\text{C}$
 - $\sigma \sim 0.5 \text{ b}$ (E=2 MeV)
- Capture reaction (Electromagnetic interaction)
 - $^3\text{He}(\alpha, \gamma)^7\text{Be}$
 - $\sigma \sim 10^{-6} \text{ b}$ (E=2 MeV)
- Weak process (Weak interaction)
 - $p(p, e^+ \nu) d$
 - $\sigma \sim 10^{-20} \text{ b}$ (E=2 MeV)



Different time scale

pp chain (I)

- $p(p, e^+ \nu) d$ reaction determines the lifetime of the sun



Nucleus with Different Time Scales

- Time period of nucleonic Fermi motion

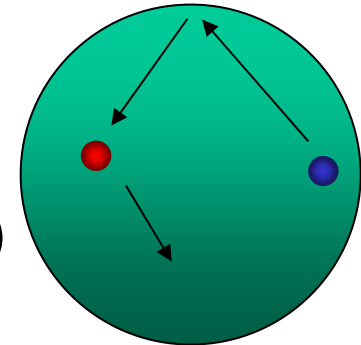
- $\tau_F \sim R/v_F \sim 10^{-22}$ sec

- Collision time

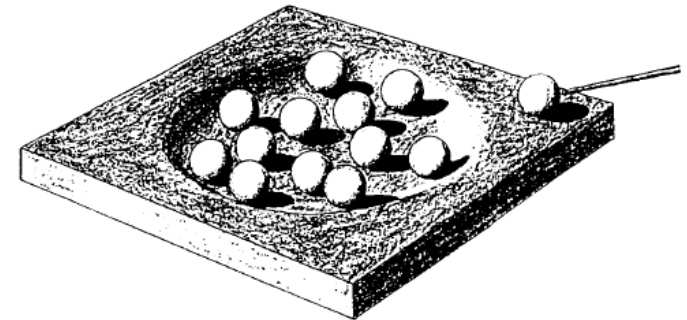
- $\tau_c \gg \tau_F$ (Nucleon near Fermi energy)

- $\tau_c \geq \tau_F$ (Thermal neutron)

- If the residence time is much larger than τ_c ,
“chaotic state” (compound nucleus).



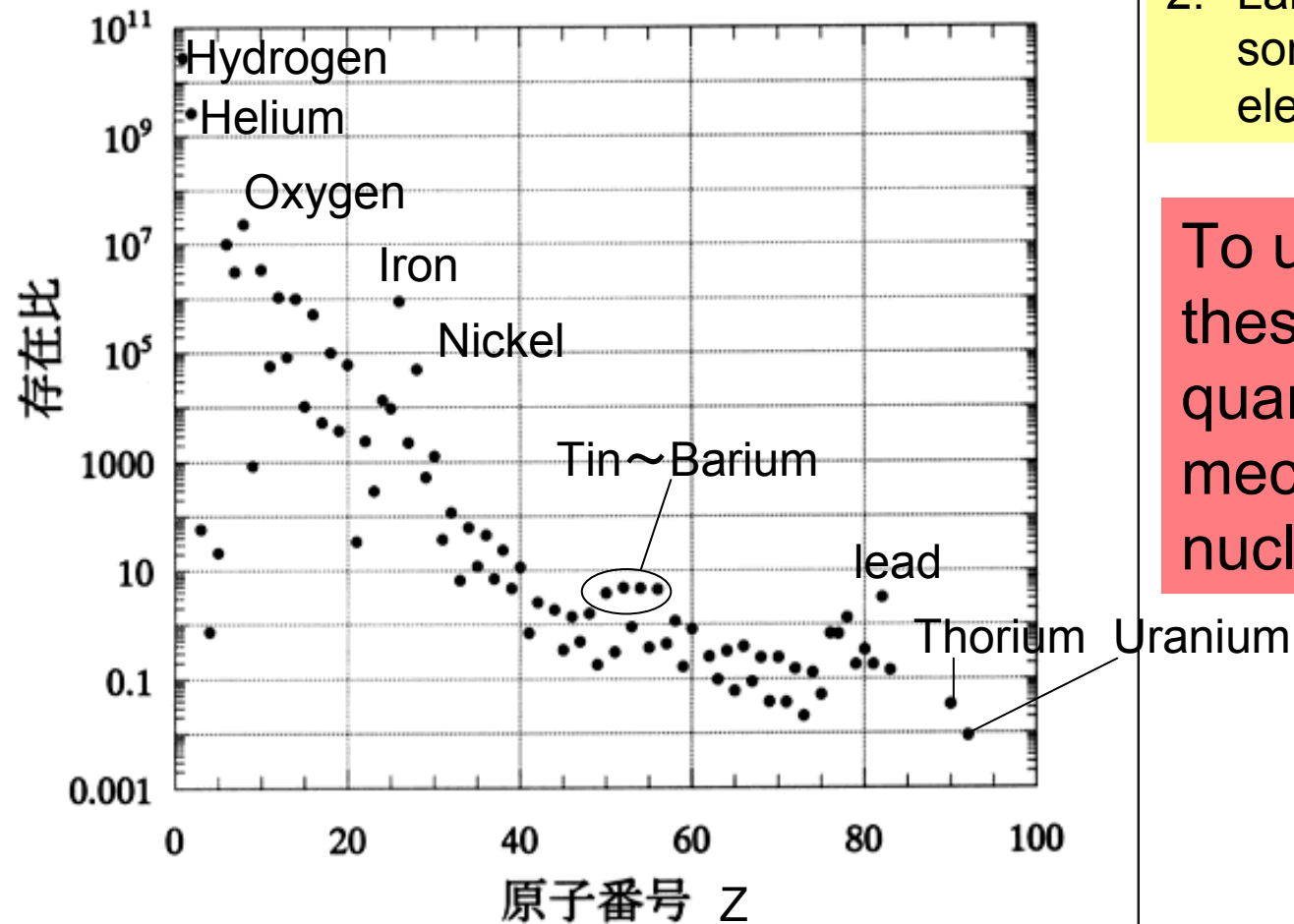
Nucleus shows different faces
(aspects) in different time scales.



Where we came from?

Solar abundance

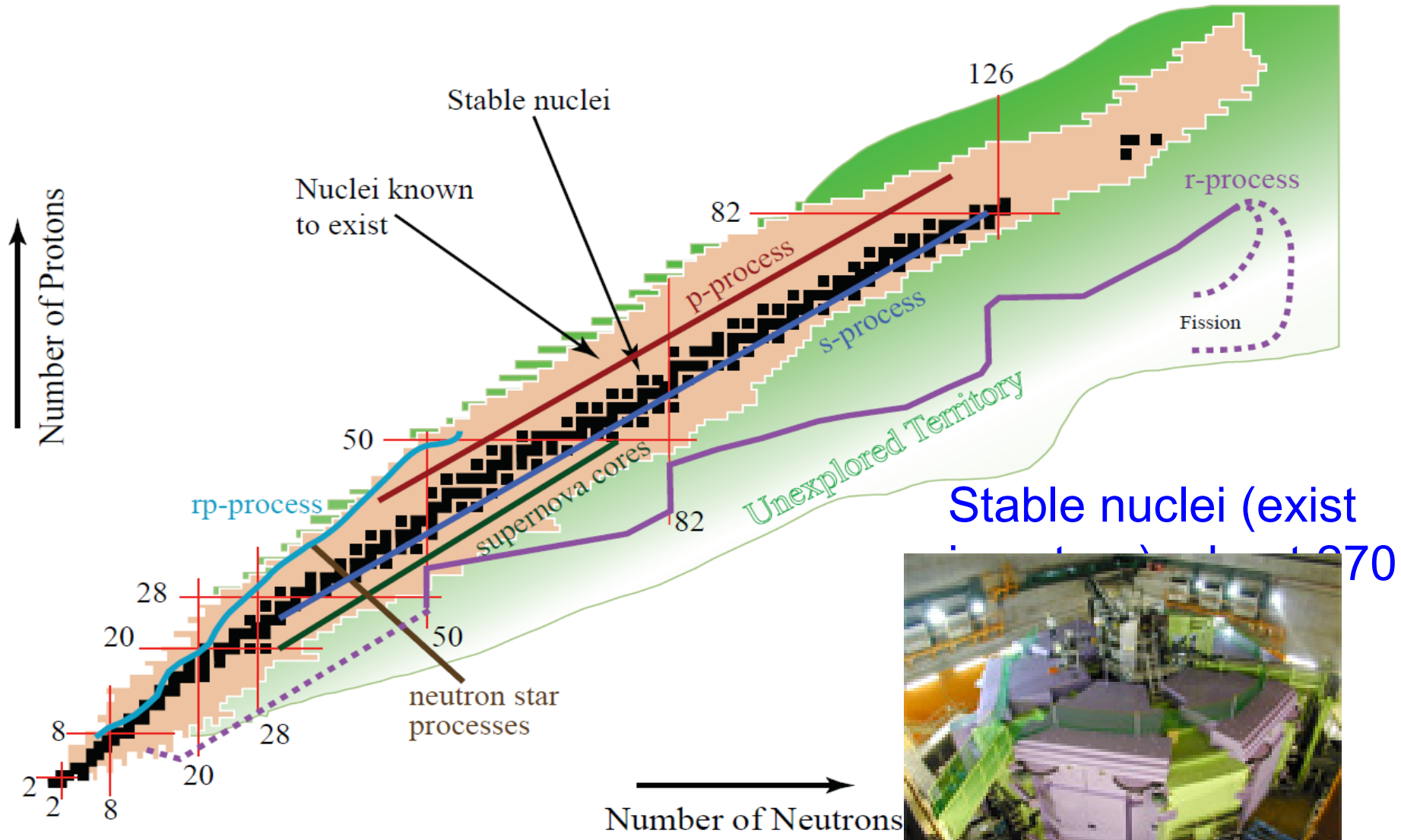
(ケイ素の元素数を 10^6 に規格化した)



1. Light nuclei > Heavy nuclei
2. Large abundance for some specific elements

To understand these, we need quantum mechanics and nuclear physics

Nuclear Chart and Element Synthesis

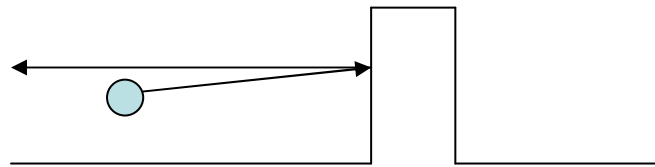
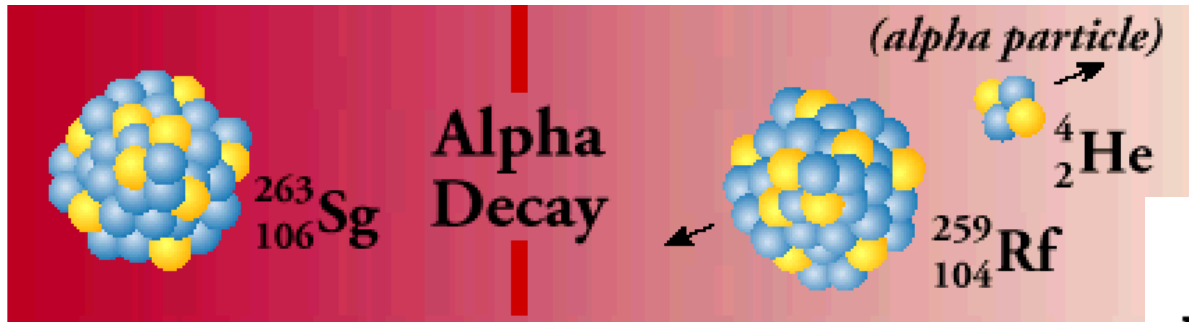


Stable nuclei (exist
until 1970)

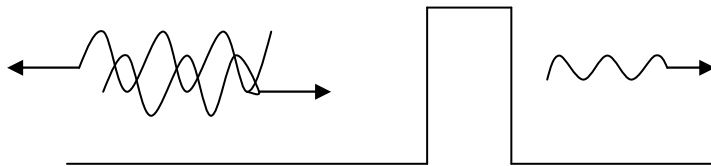


RIKEN Superconducting Cyclotron

Tunnel effect (alpha decay)

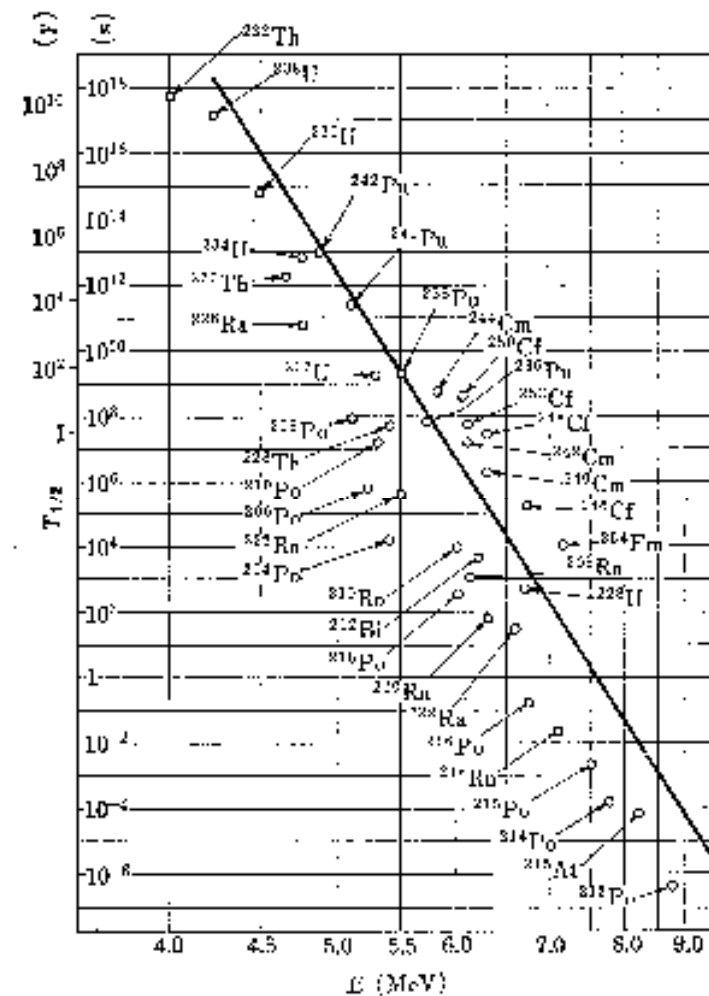


Quantum tunneling phenomena



Geiger-Nuttal $\log T_{1/2} = \frac{a}{\sqrt{E}} + b$

George Gamow uses the quantum mechanics to show the rule (1928).



Nuclear Power



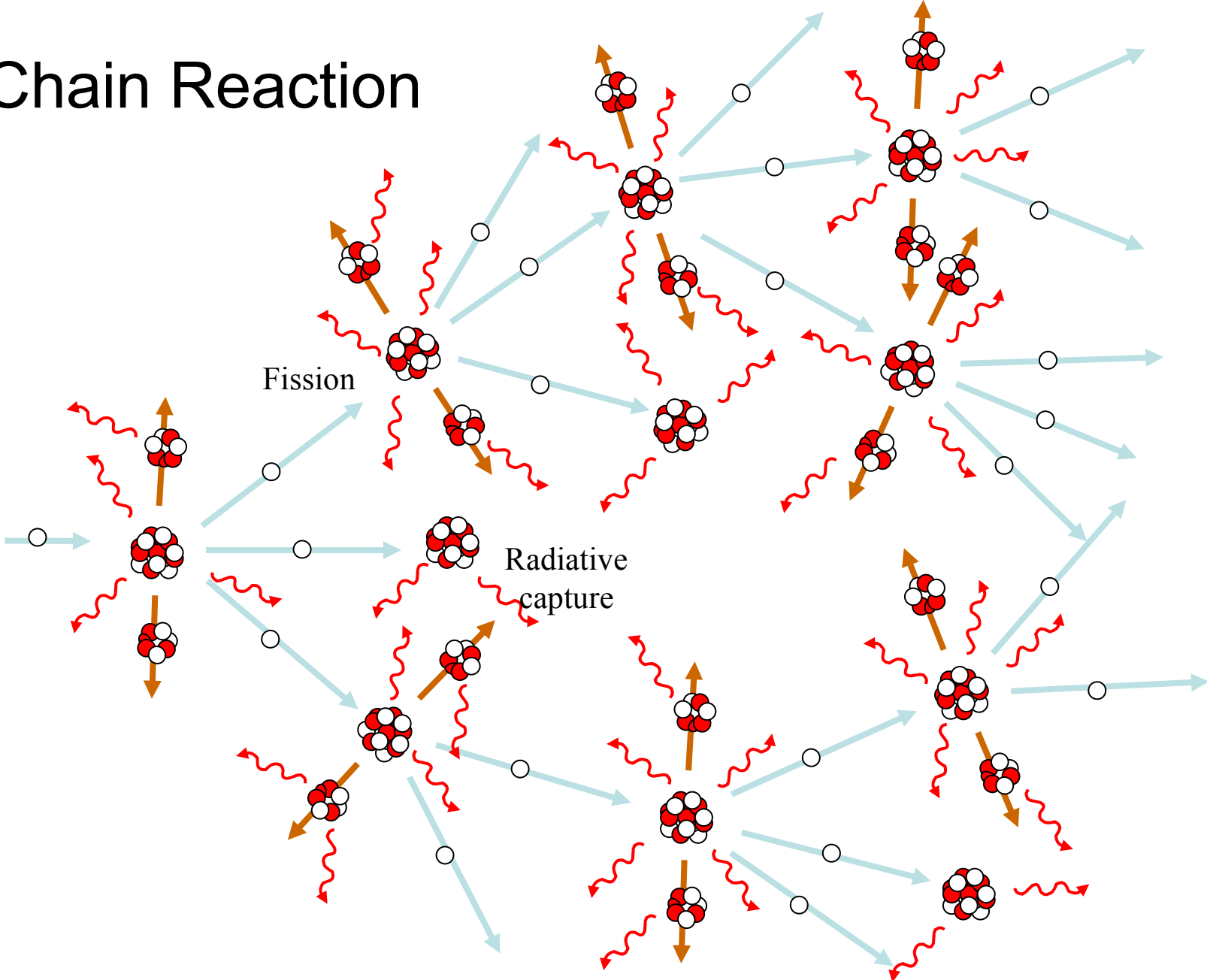
If Uranium absorbs a slow neutron, the fission occurs and neutrons are emitted. These neutrons are decelerated and collide with other Uranium.

(Chain Reaction)

The neutron absorption cross section of ^{235}U enhances at low energy (~several thousands barns).

The chain reaction stops if the speed of neutrons are too fast.

Chain Reaction



Neutron Cross Section

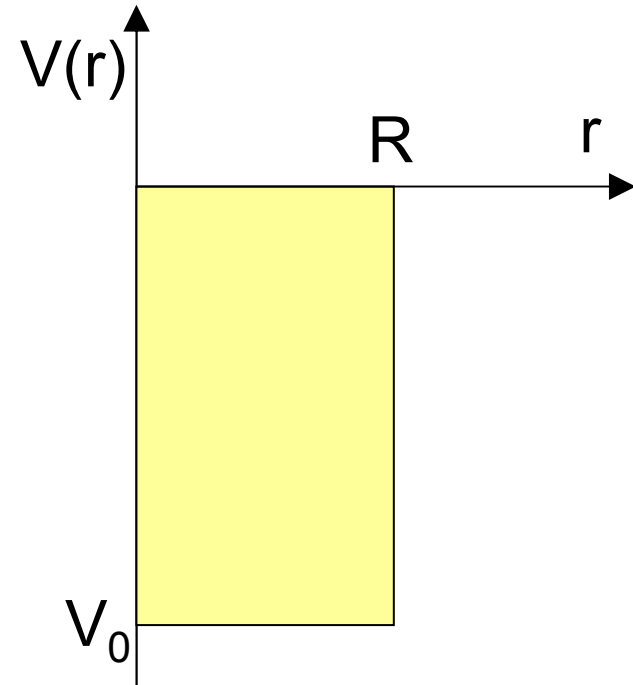
The radius of stable nucleus is approximately given by

$$R = r_0 A^{1/3}, \quad r_0 = 1.2 \text{ fm}, \quad A = N + Z$$

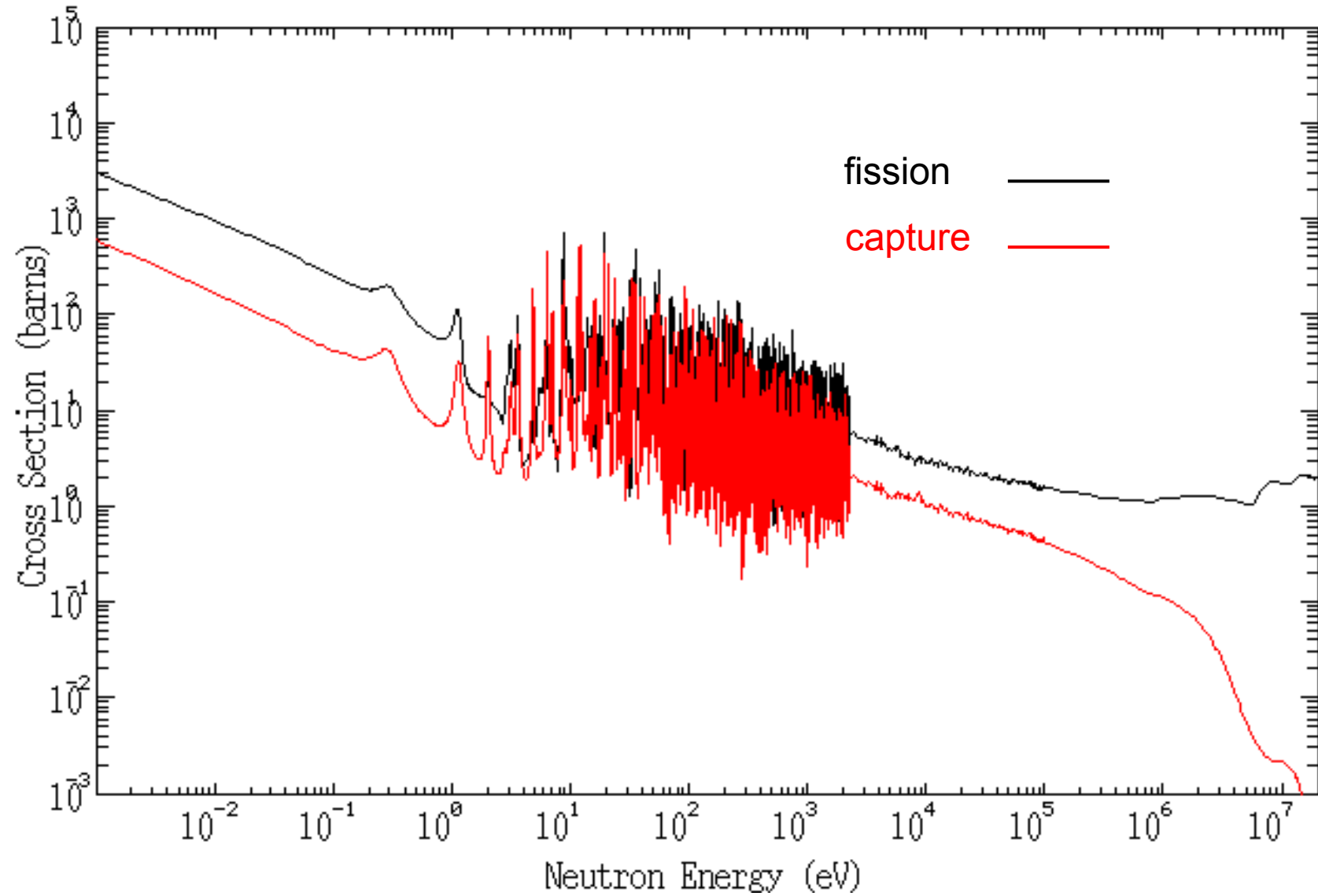
Assume a simple square-well potential model of radius R , obtain cross section in the classical mechanics?

In the quantum mechanics, the cross section depends on energy.

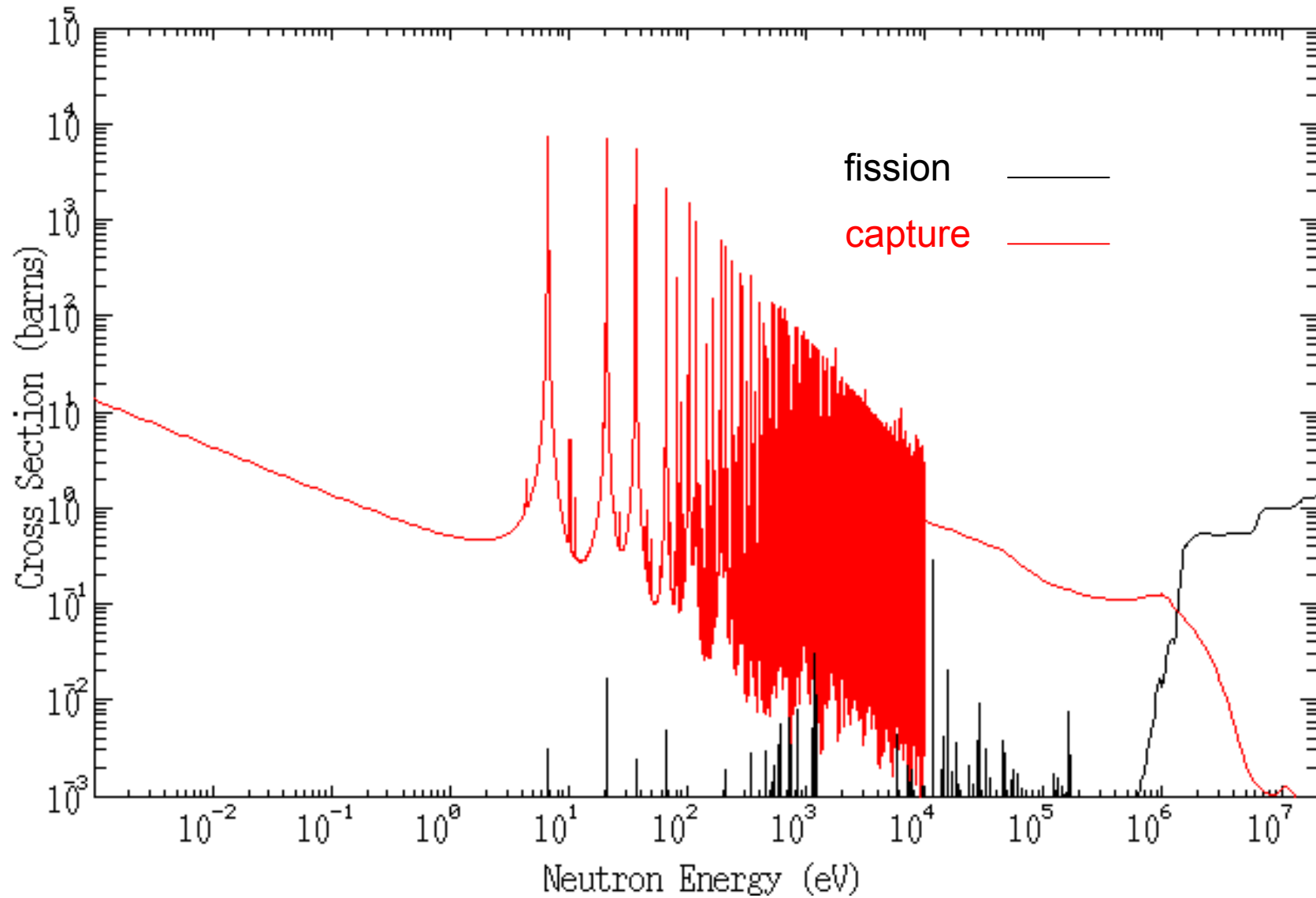
At low-energy limit,



^{235}U Cross Sections



^{238}U Cross Sections



Nuclear Landscape

- Ab initio
- Configuration Interaction
- Density Functional Theory

