

# Recent advancements in inorganic scintillator detectors

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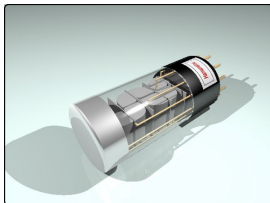
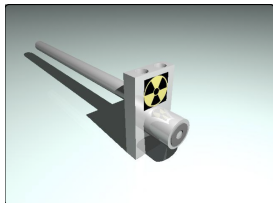
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RIKEN, Japan



1. Introduction
  - 1.1. Scintillation
  - 1.2. Applications
  - 1.3. Benchmark
2. SHOGUN
  - 2.1. LaBr<sub>3</sub> best candidat?
  - 2.2. Basic understanding
  - 2.3. LaBr<sub>3</sub> limitations
3. Recent discovery
  - 3.1. Where do we stand
  - 3.2. Pro and Con
  - 3.3. Perspectives



## Excitation sources

- Gamma-ray
- Neutrons
- Alpha...

## Samples

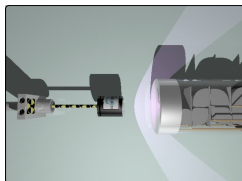
- Single crystal
- Ceramic
- Composite...

## Detectors

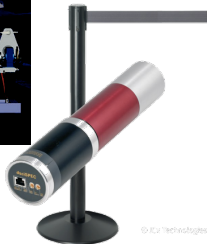
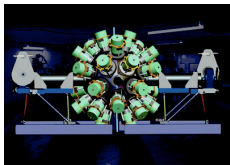
- PMT
- APD
- HybridPMT...

## Common goal

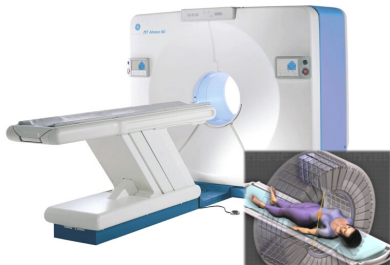
Convert the high energy of the particles or  $\gamma$ -ray into detectable light



## $\gamma$ -ray spectroscopy



## Medical imaging



## Ideal scintillator

Ideal scintillator should have all these properties:

### 1 Performance

- Absorption ⇒ Density
- Efficiency ⇒ Light output
- Speed ⇒ Decay time, Rise time
- Accuracy ⇒ Energy resolution

### 2 Cost

- Cheap ⇒ Crystal growth yield

## Ideal scintillator does not exist

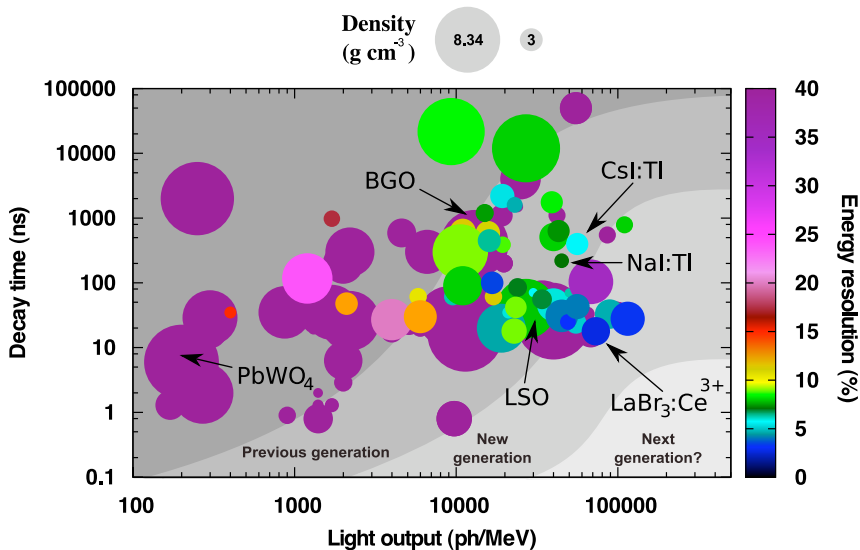
Application dependent requirements: different for High energy physics, homeland security or medical imaging applications

## Requirements

A next generation, **Scintillator-based High-resolution Gamma spectrometer for Unstable Nuclei (SHOGUN)** is proposed that is ideally suited for in-beam experiments with fast beams of rare isotopes at the RIBF with velocities of 5060% of the speed of light. The array will be based on the novel **scintillator**  $\text{LaBr}_3(\text{Ce})$ , which has exceptional properties making it the ideal material for such a spectrometer. Its **very high light output** and the **very short decay time** of the scintillation light results in an **unprecedented energy resolution** for a scintillation counter and a time resolution that is **much faster than that of NaI(Tl) or HPGe detectors**. Furthermore, due to the **large attenuation coefficient for  $\gamma$ -rays**, a compact array can achieve a very high full-energy peak efficiency.

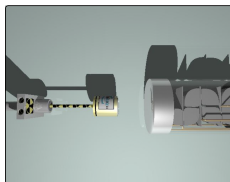
### Ideal scintillator

$\text{LaBr}_3$  scintillator?  
Can we do better?

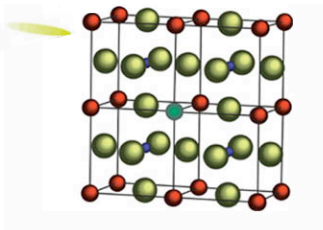


## Absorption mechanism

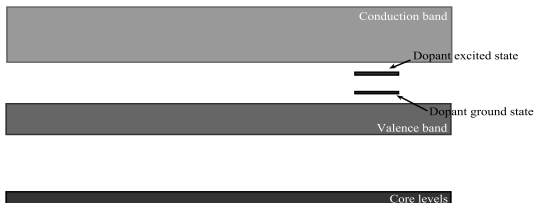
$\gamma$ -ray interacts with the lattice only



## Structural scheme



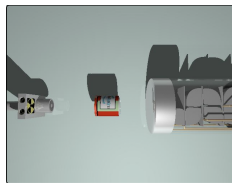
## Energy scheme



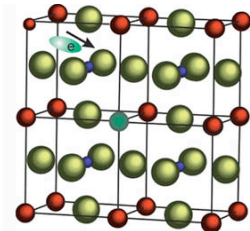


## Absorption mechanism

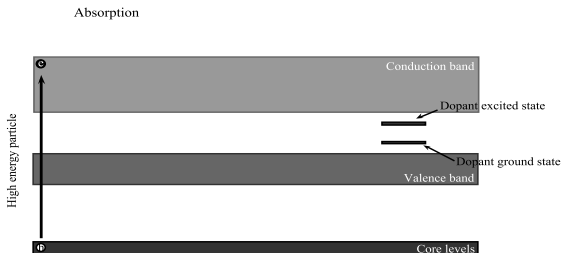
$\gamma$ -ray is converted into a high energy electron



## Structural scheme

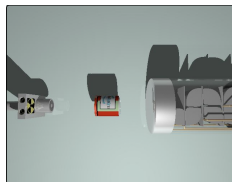


## Energy scheme

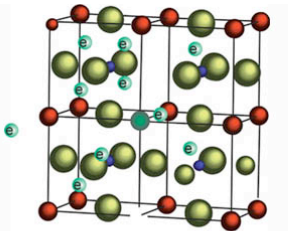


## Multiplication mechanism

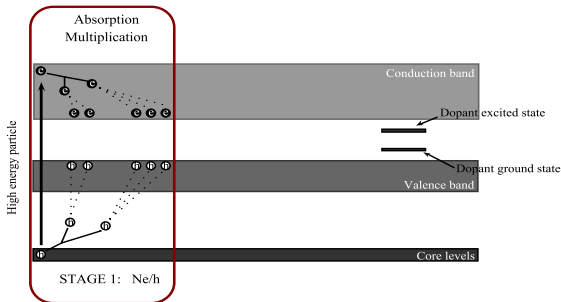
High energy electron creates e/h pairs losing its energy



## Structural scheme

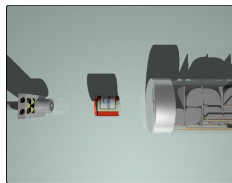


## Energy scheme

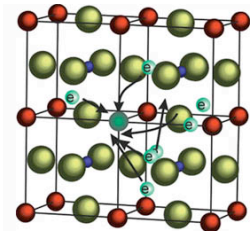


## Migration mechanism

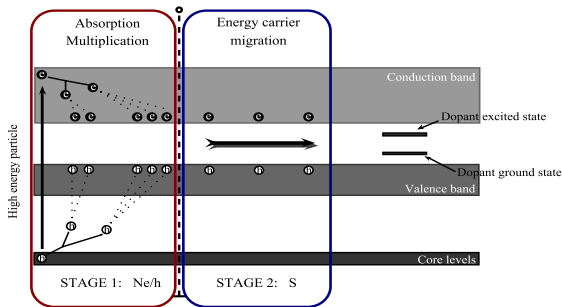
e/h pairs migrate toward the dopant ions



## Structural scheme

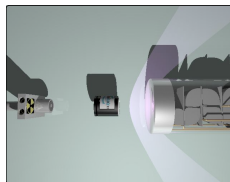


## Energy scheme

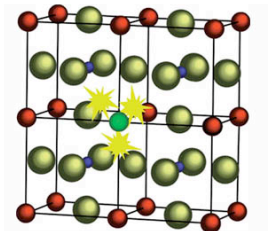


## Emission mechanism

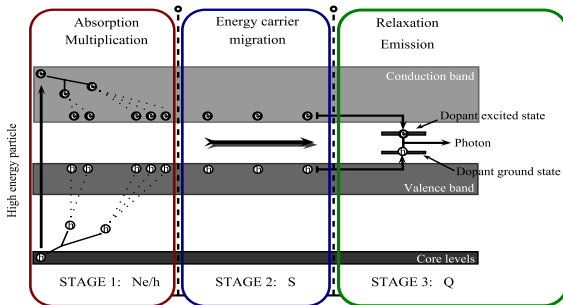
e/h pairs recombine on the dopant ions



## Structural scheme



## Energy scheme



## Formula

$$\bullet \text{ LO} = \frac{1,000,000}{\beta E_g} \text{SQ}$$

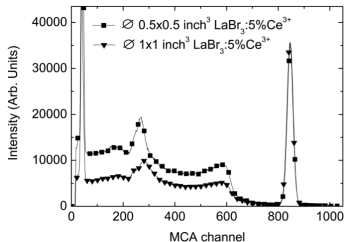
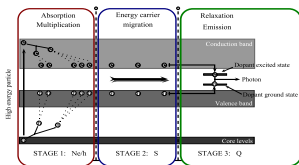
## Estimation for LaBr<sub>3</sub>

$$\bullet E_g = 6.2 \text{ eV}$$

$$\bullet \beta \sim 2$$

$$\Rightarrow \text{LO}_{\text{estimated}} = 81,000 \text{ ph/MeV}$$

$$\Rightarrow \text{LO}_{\text{measured}} = 60,000 \text{ ph/MeV}$$

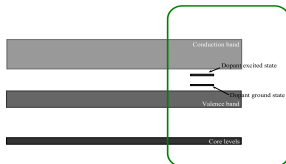


**Ideal scintillator: No**

Energy loss as large as 20% - **S or/and Q < 1**

## Quantum efficiency, $Q$

- Ce<sup>3+</sup> energy levels vs Valence and Conduction bands

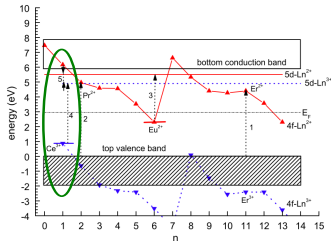


## LaBr<sub>3</sub> energy level positions

- Spectroscopy
- First principle calculation

⇒  $4f/VB \sim 0.5 \text{ eV}$

⇒  $5d/CB \sim 1 \text{ eV}$

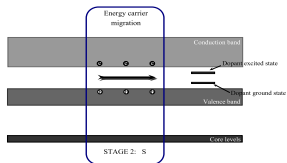


**Ideal scintillator: No**

Energy loss due to energy transfer -  $S < 1, Q = 1$

## Transfer efficiency, $S$

- Self Trapped Exciton (STE) to Ce<sup>3+</sup> transfer

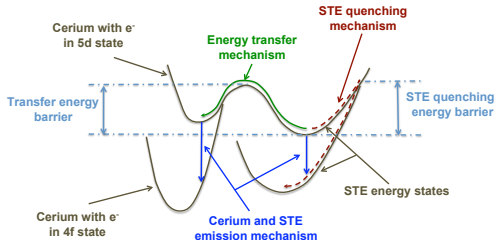


## STE properties dependence

- 2 different STEs
- 1 does not transfer entirely

⇒ Transfer is efficient

⇒ But not perfect



**Ideal scintillator: No**

Light output can be improved: better  $S$

## Formula

$$\frac{1}{\tau_{\text{Scintillation}}} = \frac{1}{\tau_{\text{Transfer}}} + \frac{1}{\tau_{\text{Dopant}}}$$

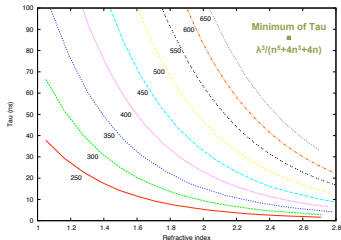
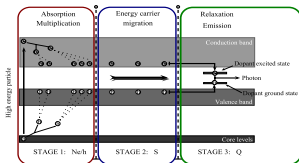
## $\tau_{\text{Dopant}}$ estimation for LaBr<sub>3</sub>

$$\tau_{\text{Ce}} \sim \frac{\lambda^3}{n^5 + 4n^3 + 4n}$$

$$n \sim 1.8$$

$$\Rightarrow \tau_{\text{estimated}} = 20 \text{ ns}$$

$$\Rightarrow \tau_{\text{measured}} = 18 \text{ ns}$$



**Ideal scintillator: Yes**

No delay due to the transfer. Ideal time response



## Formula

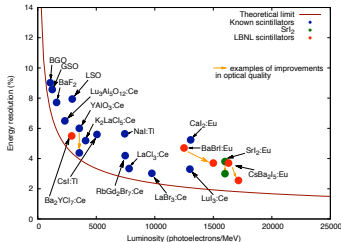
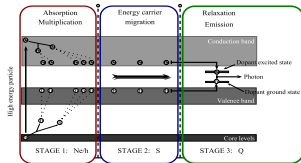
$$\bullet \text{ E.R.} = \frac{2.35}{\sqrt{\text{Number}_{\text{photoelectron}}}}$$

## E.R. estimation for LaBr<sub>3</sub>

$$\bullet \text{ Number}_{\text{photoelectron}} \sim 9,500$$

$$\Rightarrow \text{E.R.}_{\text{Estimated}} = 2.4\%$$

$$\Rightarrow \text{E.R.}_{\text{measured}} = 2.8\%$$



**Ideal scintillator: Yes**

Ideal energy resolution at 662 keV

## E.R. vs Energy

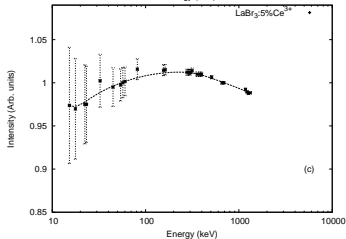
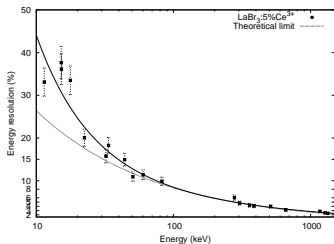
- Degradation of E.R. under 50 keV

$$\Rightarrow \text{E.R.}_{\text{measured}}^{662\text{keV}} = \text{E.R.}_{\text{theoretical}}^{662\text{keV}}$$

$$\Rightarrow \text{E.R.}_{\text{measured}}^{10\text{keV}} = 1.5 \text{ E.R.}_{\text{theoretical}}^{10\text{keV}}$$

## Non proportionality

- E.R. increases with increasing non proportionality



**Ideal scintillator: Yes**

Deviation of theoretical energy resolution under 50 keV

## <sup>138</sup>La $\gamma$ -ray background

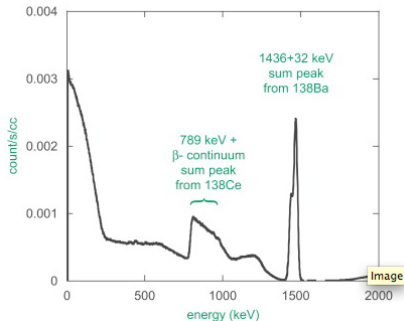
⇒ Lanthanum  $\in$  0.09% <sup>138</sup>La

⇒ Half-life of 1.05E<sup>11</sup> years

⇒ 1.5 Bq/cc in LaBr<sub>3</sub>

## Energy spectrum

- <sup>138</sup>Ba by electron capture:  
Gamma-ray at 1436 keV + 32 keV Ba x-ray
- <sup>138</sup>Ce by  $\beta$ -emission:  
Gamma-ray at 789 keV

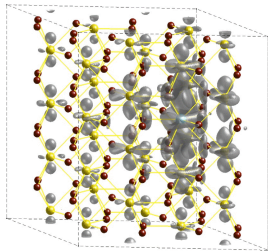


**Ideal scintillator: No**

Application relevancy

## Structural parameter

- ⇒ Density [g/cm<sup>3</sup>] = 5.08
- ⇒ Melting point [K] = 1116
- ⇒ Thermal expansion coefficient [10<sup>-6</sup>/C] = 8 along C-axis
- ⇒ Cleavage plane: <100>
- ⇒ Hygroscopic: High



**Ideal scintillator: No**

Crystal growth yield is low.

## Pro

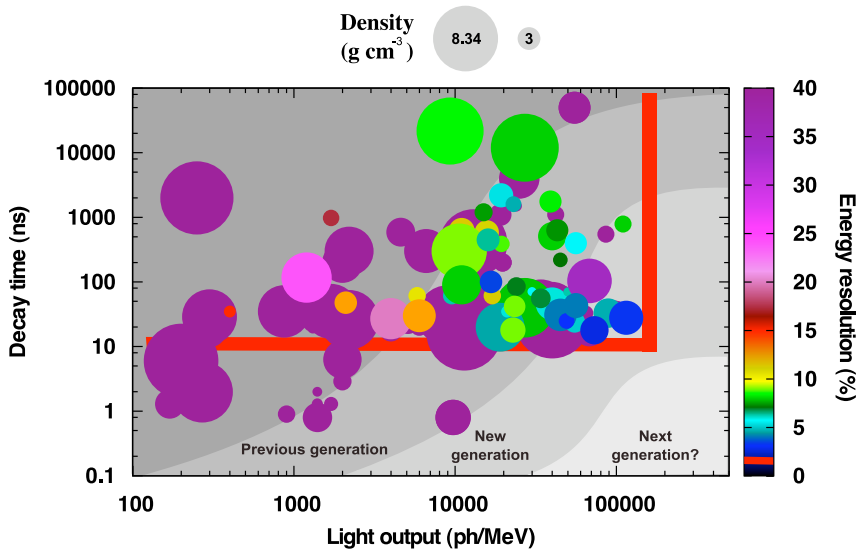
- **Time response:** 18 ns and Time resolution: 0.450 ns
- **Energy resolution:** 2.8% at 662 keV
- **Light output:** 60,000 ph/MeV

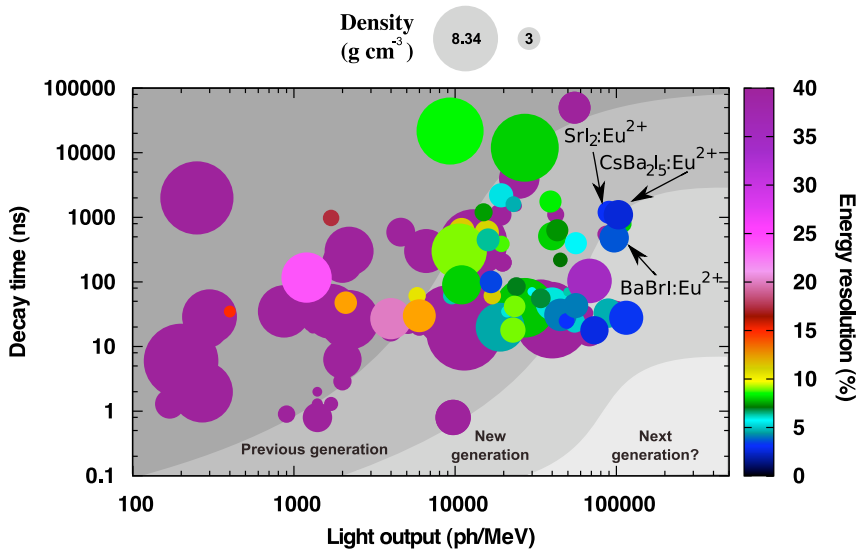
## Con

- **Cost:** thermal anisotropy leading to low growth yield
- **Light output:** substantial energy loss
- **Energy resolution:** Degradation in the low energy range
- **Hygroscopicity:** High
- **Intrinsic radioactivity:** <sup>138</sup>La  $\gamma$ -ray background

Yes but...

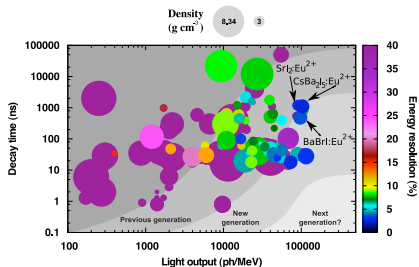
Can we do better?





## Recent development

- CLLB
- $\text{SrI}_2$
- $\text{CsBa}_2\text{I}_5$  and  $\text{BaBrI}$



## Comparatif

Scintillator	$\text{NaI:Tl}$	$\text{LaBr}_3:\text{Ce}^{3+}$	$\text{CLLB}:\text{Ce}^{3+}$	$\text{SrI}_2:\text{Eu}^{2+}$	$\text{CsBa}_2\text{I}_5:\text{Eu}^{2+}$	$\text{BaBrI}:\text{Eu}^{2+}$
Development (year)	60	10	4?	4?	1	1
Structure	Cubic	Hexa	Cubic	Ortho	Mono	Ortho
Band gap (est. in eV)	5.9	6.2	5.8	5.4	5.1	5.3
Density ( $\text{g cm}^{-3}$ )	3.67	5.1	4.2	4.5	5.0	5.0
Decay time (ns)	230	17	55	1,200	1,400	500
Luminosity (ph/MeV)	42,000	60,000	50,000	100,000	103,000	91,000
Energy resolution (%)	6.5	2.8	3.0	3.0	2.55	3.3
Detection efficiency (%)	30	26	-	31	36	35
Hygroscopicity	High	High	-	High	Low	Low



## Criteria

- **Band gap**

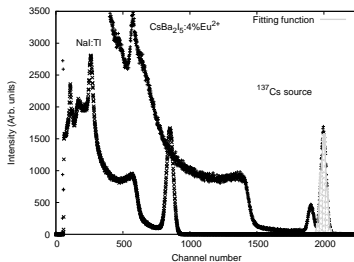
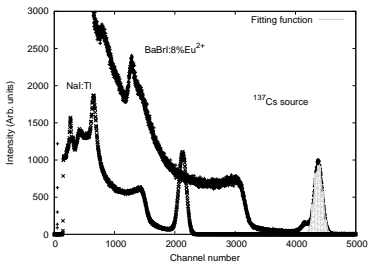
BaBrI:Eu<sup>2+</sup> ~ 5.3 eV, CsBa<sub>2</sub>I<sub>5</sub>:Eu<sup>2+</sup> ~ 5.1 eV

- **Estimated Light output**

BaBrI:Eu<sup>2+</sup> ~ 95,000 ph/MeV, CsBa<sub>2</sub>I<sub>5</sub>:Eu<sup>2+</sup> ~ 100,000 ph/MeV

- **Measured Light output**

BaBrI:Eu<sup>2+</sup> ~ 91,000 ph/MeV, CsBa<sub>2</sub>I<sub>5</sub>:Eu<sup>2+</sup> ~ 100,000 ph/MeV



## LaBr<sub>3</sub> vs BBI, CBI

Smaller band gap and better transfer efficiency

## Gamma-ray response

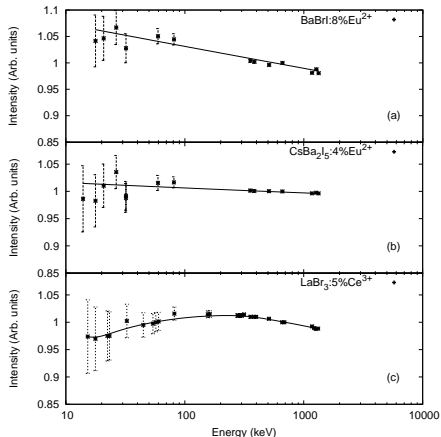
### Benchmark

$$\sigma_{np} = \sqrt{\frac{1}{N} \sum_{i=1}^N \left( \frac{Y(E_i)}{Y(^{137}\text{Cs})} - 1 \right)^2}$$

$$\Rightarrow \sigma_{np}(\text{BaBrI}) = 0.024$$

$$\Rightarrow \sigma_{np}(\text{CsBa}_2\text{I}_5) = 0.009$$

$$\Rightarrow \sigma_{np}(\text{LaBr}_3) = 0.013$$



**LaBr<sub>3</sub> vs BBI, CBI**

Better response for CBI

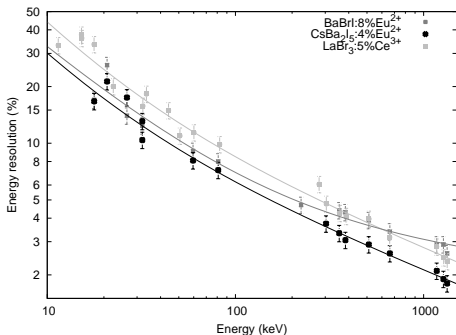
## Energy resolution at 662 keV

$$\Rightarrow \text{E.R.}_{\text{BaBrI}}^{662\text{keV}} = 3.3\%$$

$$\Rightarrow \text{E.R.}_{\text{CsBa2I5}}^{662\text{keV}} = 2.55\%$$

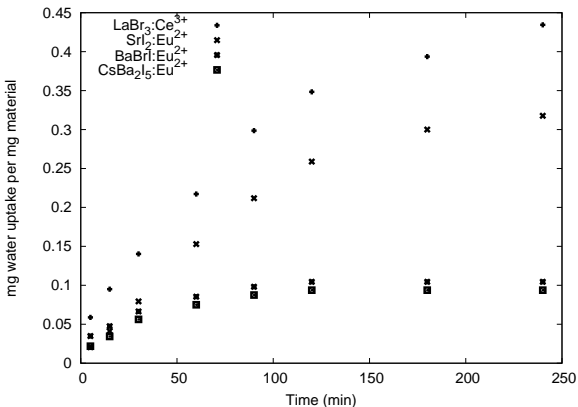
## Energy resolution vs Energy

- Both scintillators give better E.R. under 50 keV



## LaBr<sub>3</sub> vs BBI, CBI

Better energy resolution for CBI and BBI under 50 keV



## LaBr<sub>3</sub> vs BBI, CBI

Hygroscopicity is less in both cases

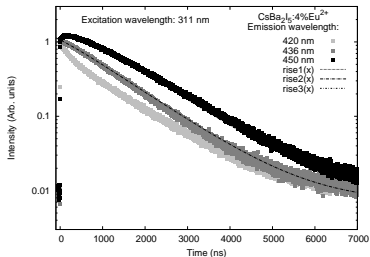
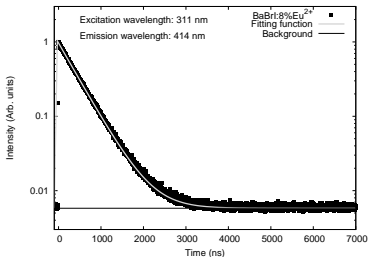
## Criteria

- Time response

BaBrI:Eu<sup>2+</sup> ~ 450 ns, CsBa<sub>2</sub>I<sub>5</sub>:Eu<sup>2+</sup> ~ 1,200 ns

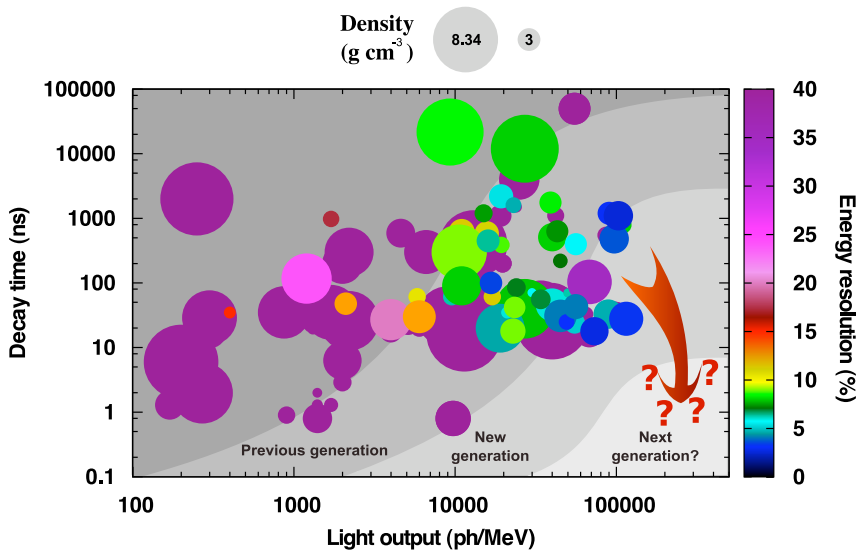
- Self absorption

BaBrI:Eu<sup>2+</sup> ~ Not detected, CsBa<sub>2</sub>I<sub>5</sub>:Eu<sup>2+</sup> ~ Medium



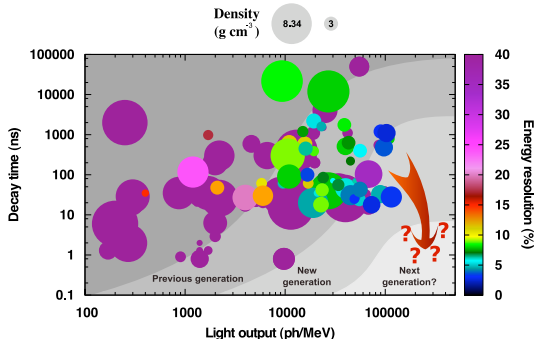
## LaBr<sub>3</sub> vs BBI, CBI

Slower than LaBr<sub>3</sub> but excellent for Eu<sup>2+</sup> doped compound



## Problems

- Limits are intrinsic to doped inorganic insulator
- Dopant intrinsic time response limits the decay time
- Dopant energy levels limit the band gap range available



## Approaches

Description	Application	Advantages	Draw back
Semi conductor scintillators	Cryogeny	Fast	Thermal stability
Nano and composite scintillators	Any	Fast, low cost	How to absorb energy?

## Fast moving research field

- **Large improvement within the past 5/10 year**
  - Understanding the physics of scintillators
  - Application via the improvement and discovery of high performance scintillators
  - Exploration of new concepts in solid state detectors

## Multidisciplinary collaboration

- Computational science
- Chemistry
- Physics
- Crystal growth
- Instrumentation...



## Berkeley National Lab

Edith Bourret, Steve Derenzo, William Moses, Seng Choong, Martin Boswell, Steve Hanrahan, Martin Janecek, Christopher Ramsey, James Powell, David Wilson, Andrew Canning, Marv Weber, Yetta Eagleman, Gautam Gundiah, Ramesh Borade, Zewu Yan, Kathleen Brennan

## Livermore National Lab

John Valentine, Stephen Payne, Nerine Cherepy

## Wake Forest University

Richard Williams

## Charles Darwin University

Jay Singh

## Moscow State University

Andrei Vasiliev

## Delft University

Pieter Dorenbos

## S<sup>t</sup> Gobain

Eric Mattman

## RMD, Inc.

Kanai Shah

## Institute Single Crystals

Alex Gektin

## University of Tennessee

Chuck Melcher