

# Basic of Detector

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# What I worked for detectors

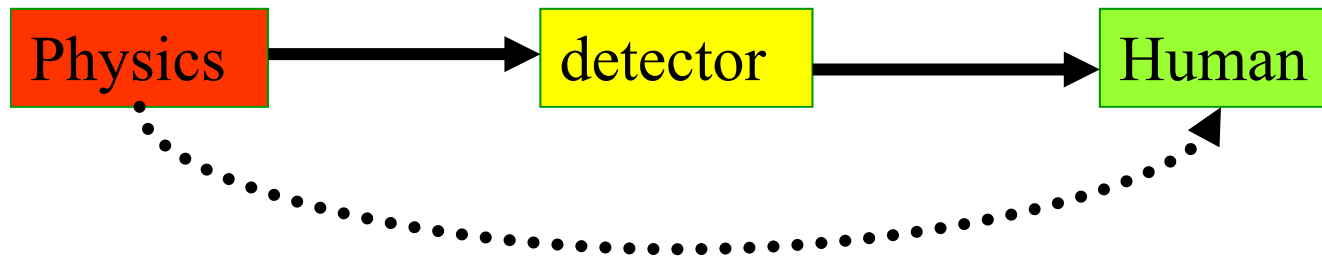
- Electron-Positron collider Experiment at 60GeV
  - Trigger electronics, TRD, EMCAL, Si Sensor
- Proton-Antiproton collider experiment at 1.8TeV
  - Muon detector, Readout electronics
- Large scale Accelerator control at 8GeV
  - Distributed computing system hardware/software
- Polarize proton-proton/ Heavy Ion collider experiment at 200GeV
  - Muon detector, Si detector
- Start to work for Detectors for RIBF experiment

Working higher energy

# Index of this lecture

1. Why/How we need detector?
2. What do we want to measure?
3. Gas Chamber basics
4. Scintillator
5. PHENIX experiment and Silicon Detector
6. Summary

# Importance of Detector



- We need detector to understand physics
- Detector innovation can arise new physics
  - Telescope (1590) : Newton mechanics (Late 1600's)
  - Velocity of light measurement (1873): Relativity (1905, 1916)
  - High resolution hydrogen spectroscopy : Quantum mechanics ( 1925)

# Discovery of Charm Quark

EW LETTERS

2 DECEMBER 1974

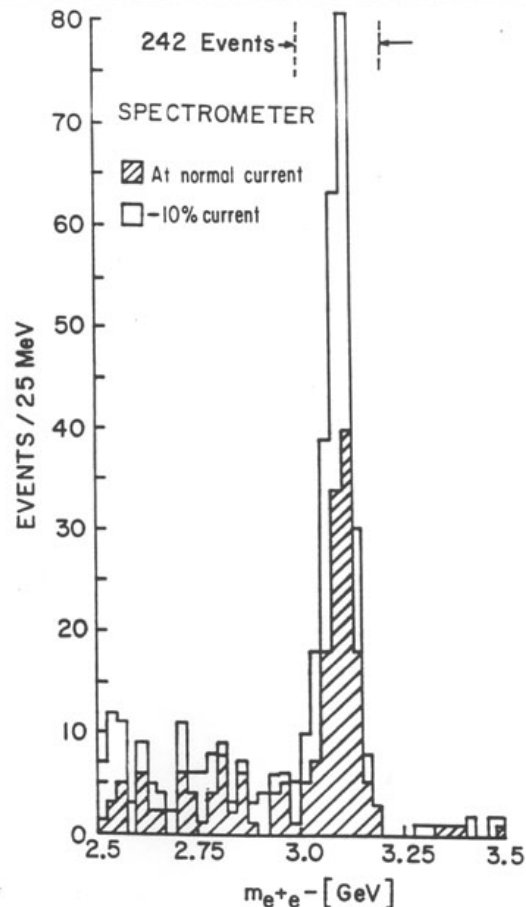
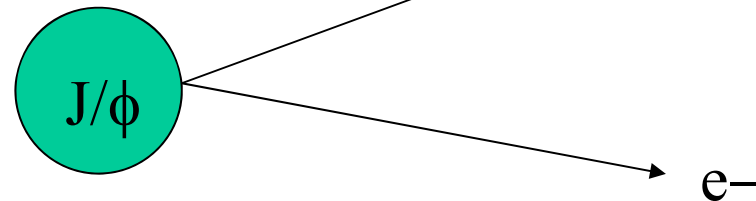


FIG. 2. Mass spectrum showing the existence of  $J/\psi$ . Results from two spectrometer settings are plotted showing that the peak is independent of spectrometer currents. The run at reduced current was taken two months later than the normal run.

On 1974 丁肇中 and B. Richter discovered independently.

Novel prize in Physics in 1976. Ting:

Energetic proton was bombarded to nuclear target, measure the invariant mass of produced electron and positron



Until Ting's discovery, many experiments saw the sign of the similar phenomena, But their resolution of the mass measurement were not good as Ting's experiment.

# Major Detector Principle

1. Particle penetrates or stops at detector

2. Particle interacts with material of detector

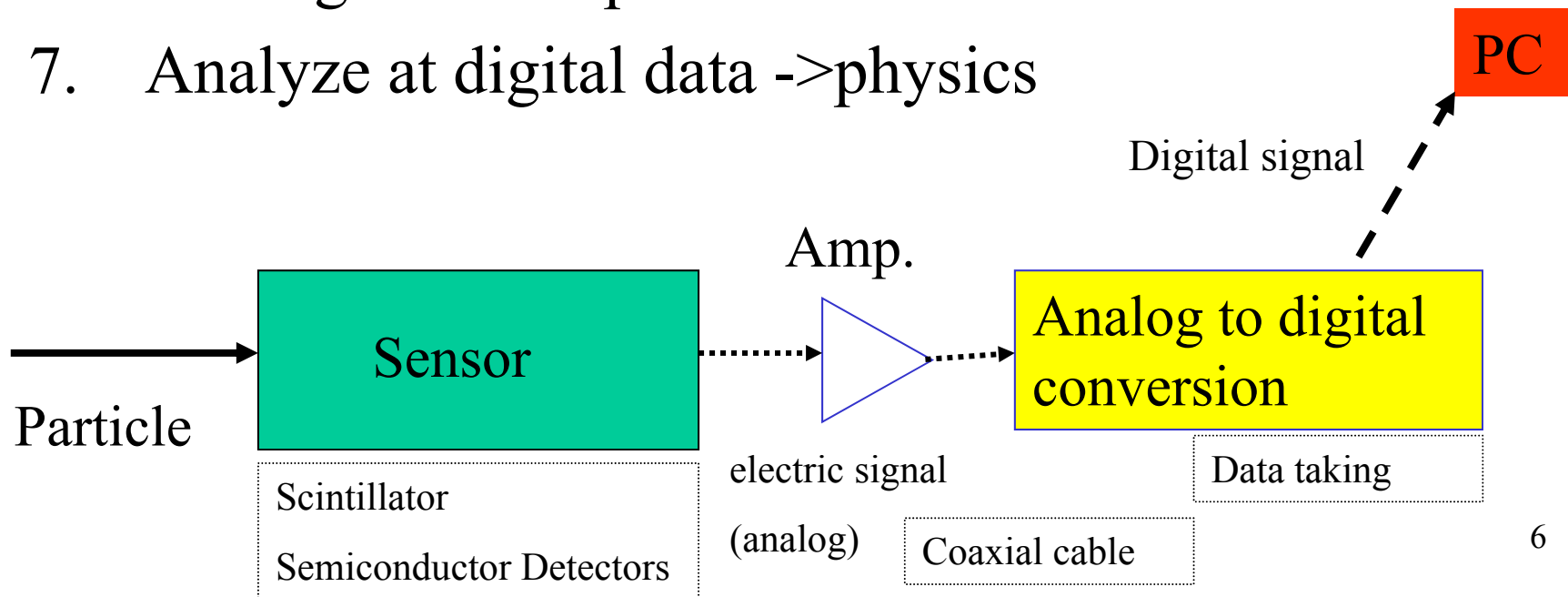
3. Generating some signal

4. Amplification mechanism

5. Analog to Digital conversion

6. Getting into computer

7. Analyze at digital data ->physics



# Particle mass

- Particle has its own mass.
  - electron  $0.511\text{MeV}$
  - $\mu$   $105\text{MeV}$
  - $\pi^+$ ,  $\pi^-$   $140\text{MeV}$ ,  $\pi^0$   $135\text{MeV}$
  - Proton  $938\text{MeV}$
  - $J/\psi$   $3097\text{MeV}$  (discovered by S.C.C. Ting and B. Richter)
  - **Top quark  $172\text{GeV}$  (heaviest particle ever observed)**
- If we know the mass of particle, we can identify the particle species.

# 4-momentum

- Treating the mass at relativity

$$P^2 = E^2 - |\mathbf{p}|^2 = m^2$$

Where  $P$  : 4-momentum

$E$  : Energy

$\mathbf{p}$  : 3-momentum (  $p_x, p_y, p_z$  )

$m$  : invariant mass



# 4-momentum

- $(E, p_x, p_y, p_z)$
- Invariant mass :  $m^2 = E^2 - |\mathbf{p}|^2$
- 3-momentum (velocity)  $\mathbf{v}/c = \boldsymbol{\beta} = \mathbf{p}/E$

Where  $c$  is light velocity

- If we can measure 3-momentum  $\mathbf{p}$ , and Energy  $E$  or 3-momentum  $\boldsymbol{\beta}$ ,  $m$  can be obtained  $\rightarrow$  identifying particle.

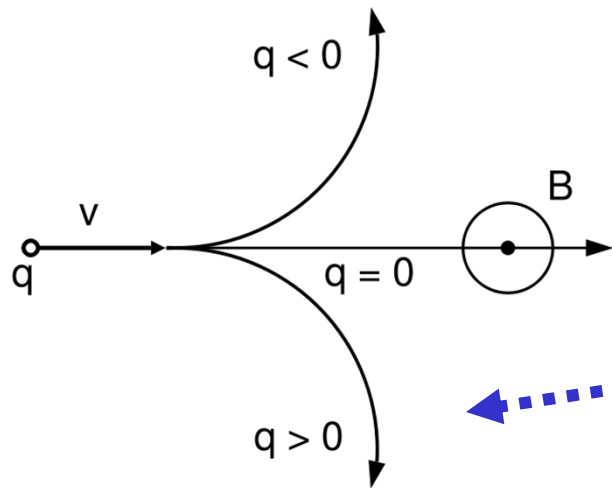
# 3-momentum measurement

- Momentum can be measured by using Lorentz force

$$\mathbf{F} = q[\mathbf{E} + (\mathbf{v} \times \mathbf{B})],$$

$\mathbf{F}$ : force,  $q$ : electric charge,  $\mathbf{E}$ : electric field

$\mathbf{v}$ : particle velocity =  $\mathbf{p}/m$ ,  $\mathbf{B}$ : magnetic field



Constant force

-> Constant curvature

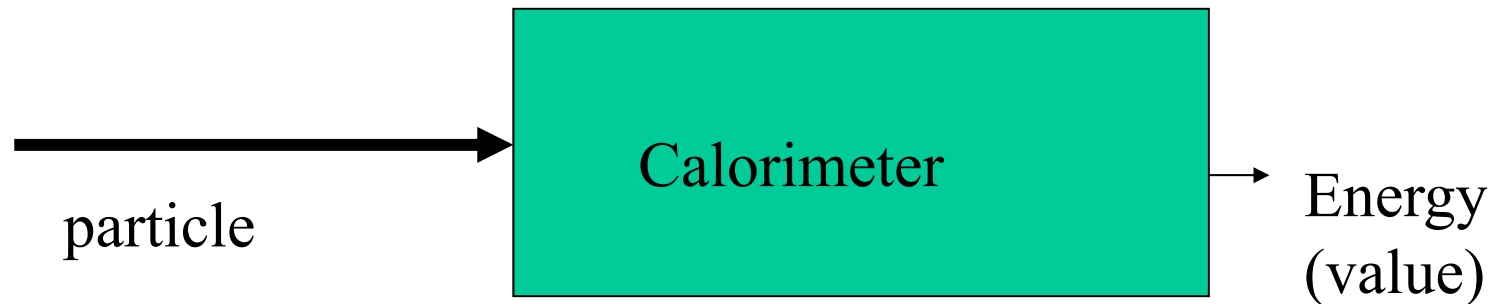
-> particle track trajectory

$$P [\text{MeV}/c] = 3 * r[\text{cm}] * B[\text{T}]$$

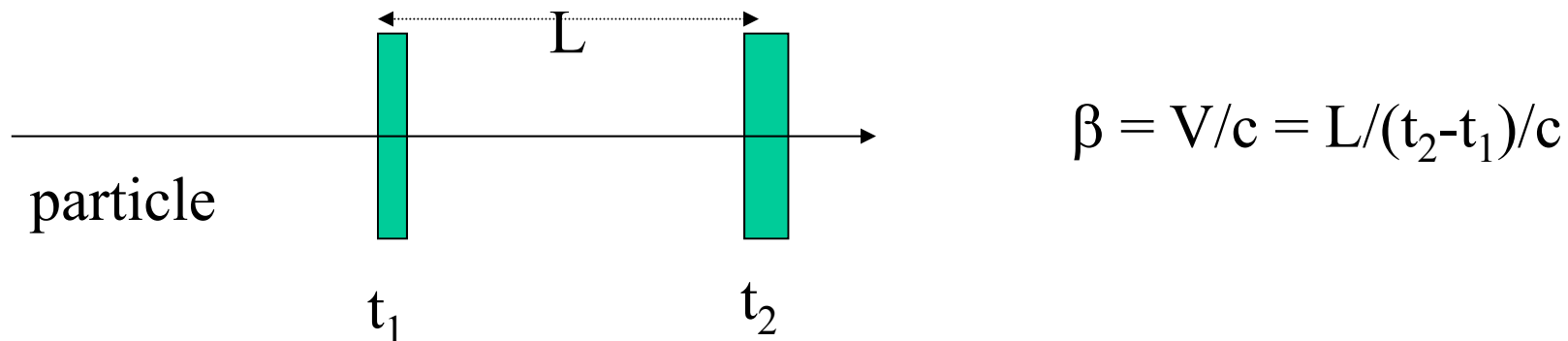
$r$ : curvature radius,  $B$ : magnetic field<sup>10</sup>

# Energy measurement

- Particle stops at the material and measure all deposited energy by energy loss



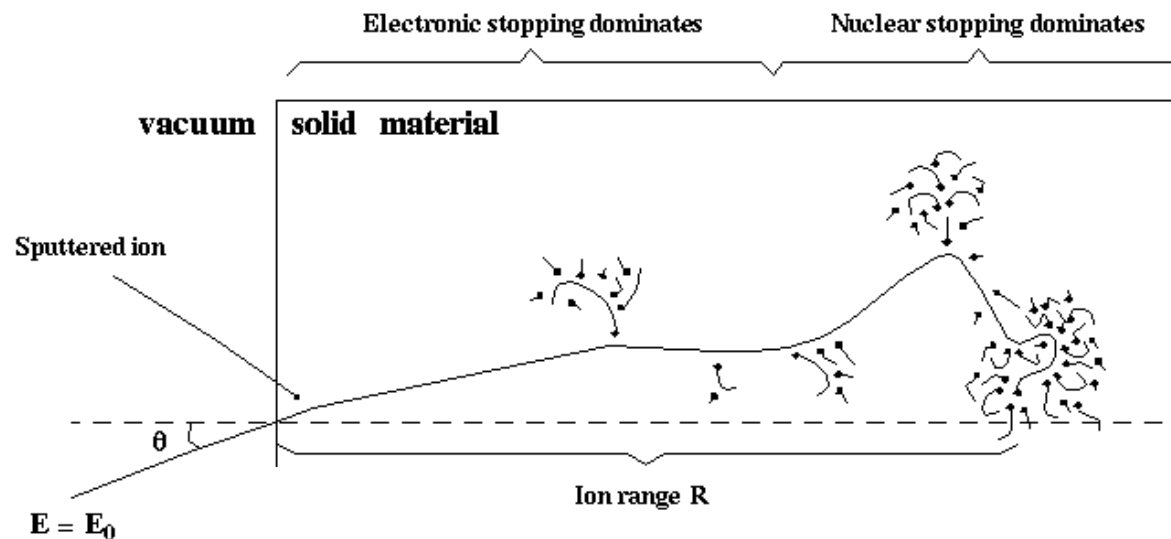
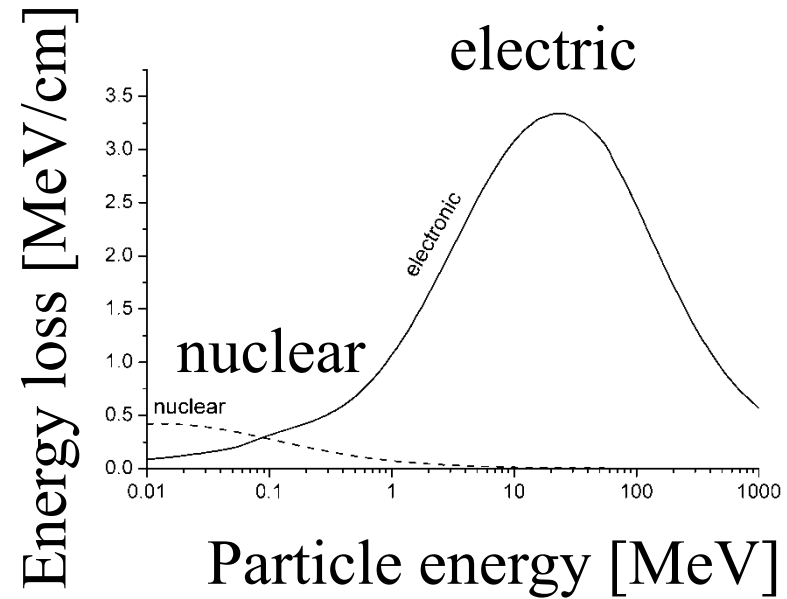
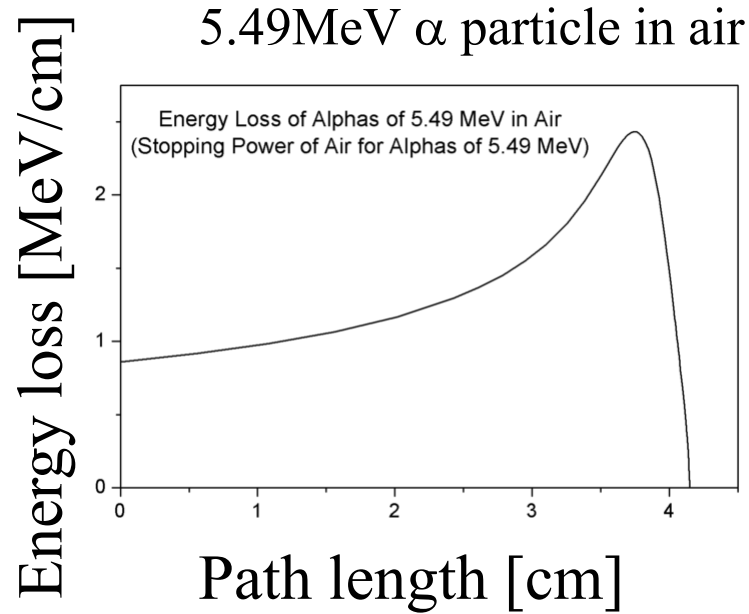
- $\beta = \mathbf{p}/E$  : particle velocity  
measure timing deference between 2  
known location



# Particle and material interaction

- Particle will hit, penetrate, or stop at material, including gas, liquid, solid.
- Particle has some interaction with material, then we can detect it. -> Detector
- Energy Loss, Multiple Scattering and So.

# Energy Loss and stopping power



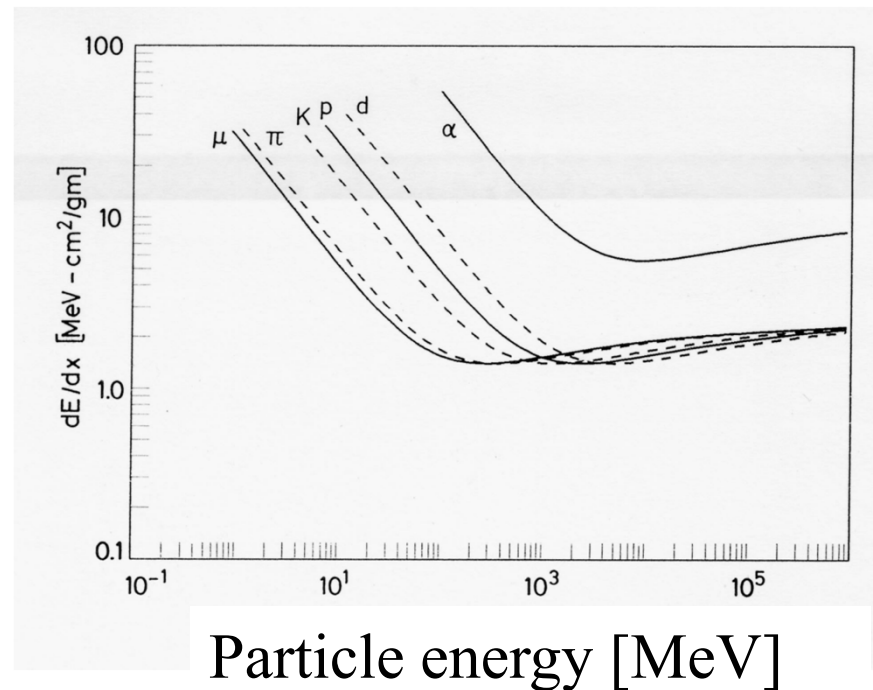
# Bethe-Bloch formula

Particle charge

$$-\frac{dE}{dx} = \frac{4\pi}{m_e c^2} \cdot \frac{n z^2}{\beta^2} \cdot \left(\frac{e^2}{4\pi\epsilon_0}\right)^2 \cdot \left[ \ln \left( \frac{2m_e c^2 \beta^2}{I \cdot (1 - \beta^2)} \right) - \beta^2 \right]$$

Energy loss  
[MeV cm<sup>2</sup>/g]

- $\beta$  =  $v / c$
- $v$  velocity of the particle
- $E$  energy of the particle
- $x$  distance travelled by the particle
- $c$  [EnergyLoss.bmp](#) speed of light
- $z$  particle charge
- $e$  charge of the [electron](#)
- $m_e$  rest mass of the electron
- $n$  electron density of the target
- $I$  mean excitation potential of the target
- $\epsilon_0$  [permittivity](#) of free space



# Typical Energy Loss

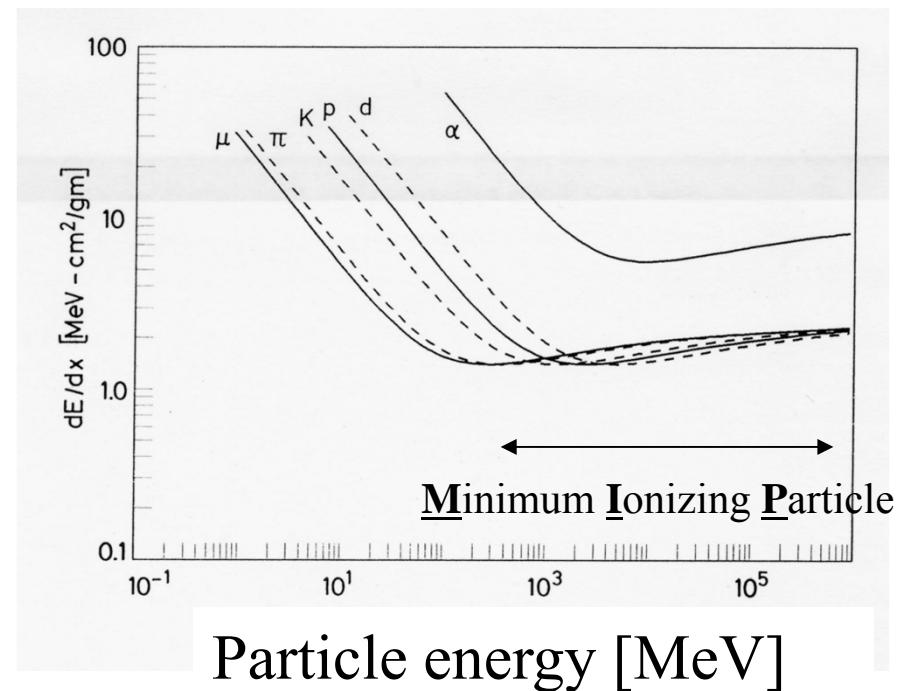
- $dE/dX \sim 1\text{MeV cm}^2/\text{g}$  for Minimum ionizing particle
- Energy loss / Unit length  
 $\sim 2\text{MeV cm}^2/\text{g} * \text{Material density} [\text{g}/\text{cm}^3]$

For example at Al

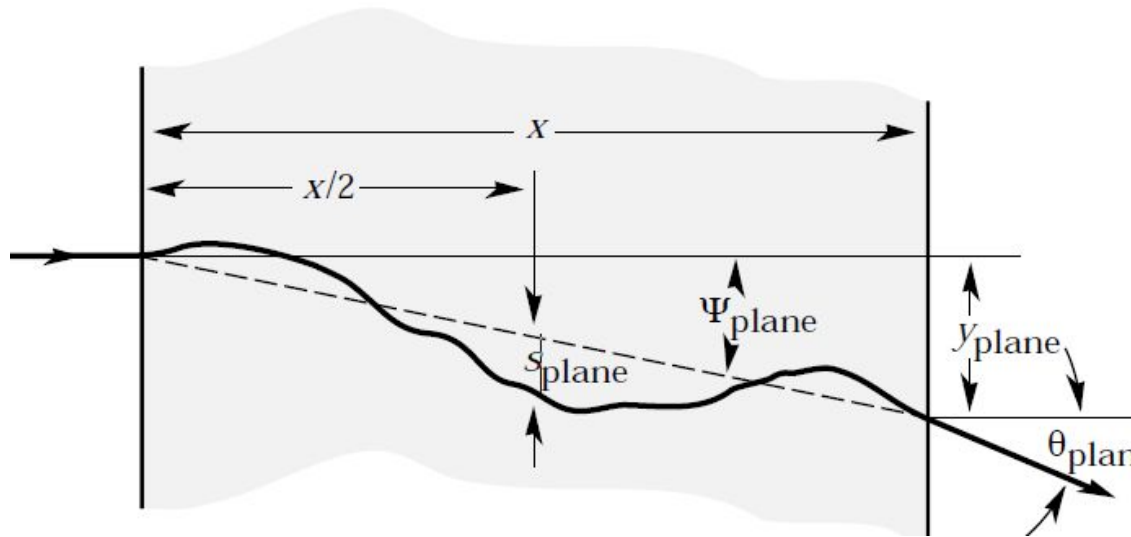
$$dE/dX = 1.615\text{MeV cm}^2/\text{g}$$

$$\text{Aluminum } \rho = 2.70\text{g}/\text{cm}^3$$

$$\text{Energy loss} = 0.60\text{MeV}/\text{cm}$$



# Multiple Scattering for M.I.P.



$$\theta_0 = \theta_{\text{plane}}^{\text{rms}} = \frac{1}{\sqrt{2}} \theta_{\text{space}}^{\text{rms}}$$

$$\psi_{\text{plane}}^{\text{rms}} = \frac{1}{\sqrt{3}} \theta_{\text{plane}}^{\text{rms}} = \frac{1}{\sqrt{3}} \theta_0 ,$$

$$y_{\text{plane}}^{\text{rms}} = \frac{1}{\sqrt{3}} x \theta_{\text{plane}}^{\text{rms}} = \frac{1}{\sqrt{3}} x \theta_0 ,$$

$$s_{\text{plane}}^{\text{rms}} = \frac{1}{4\sqrt{3}} x \theta_{\text{plane}}^{\text{rms}} = \frac{1}{4\sqrt{3}} x \theta_0$$

$$\theta_0 = \frac{13.6 \text{ MeV}}{\beta c p} z \sqrt{x/X_0} \left[ 1 + 0.038 \ln(x/X_0) \right]$$


Z: particle charge

x: material thickness

$X_0$ : radiation length



# Atomic and Nuclear properties of materials

Material	Z	A	$\langle Z/A \rangle$	Nucl.coll. length $\lambda_T$ {g cm <sup>-2</sup> }	Nucl.inter. length $\lambda_I$ {g cm <sup>-2</sup> }	Rad.len. $X_0$ {g cm <sup>-2</sup> }	$dE/dx _{\min}$ { MeV g <sup>-1</sup> cm <sup>2</sup> }	Density {g cm <sup>-3</sup> {(gℓ <sup>-1</sup> )}	Melting point (K)	Boiling point (K)	Refract. index (@ Na D)
H <sub>2</sub>	1	1.00794(7)	0.99212	42.8	52.0	63.04	(4.103)	0.071(0.084)	13.81	20.28	1.11[132.]
D <sub>2</sub>	1	2.01410177803(8)	0.49650	51.3	71.8	125.97	(2.053)	0.169(0.168)	18.7	23.65	1.11[138.]
He	2	4.002602(2)	0.49967	51.8	71.0	94.32	(1.937)	0.125(0.166)		4.220	1.02[35.0]
Li	3	6.941(2)	0.43221	52.2	71.3	82.78	1.639	0.534	453.6	1615.	
Be	4	9.012182(3)	0.44384	55.3	77.8	65.19	1.595	1.848	1560.	2744.	
C diamond	6	12.0107(8)	0.49955	59.2	85.8	42.70	1.725	3.520			2.42
C graphite	6	12.0107(8)	0.49955	59.2	85.8	42.70	1.742	2.210			
N <sub>2</sub> 	7	14.0067(2)	0.49976	61.1	89.7	37.99	(1.825)	0.807(1.165)	63.15	77.29	1.20[298.]
O <sub>2</sub>	8	15.9994(3)	0.50002	61.3	90.2	34.24	(1.801)	1.141(1.332)	54.36	90.20	1.22[271.]
F <sub>2</sub>	9	18.9984032(5)	0.47372	65.0	97.4	32.93	(1.676)	1.507(1.580)	53.53	85.03	[195.]
Ne	10	20.1797(6)	0.49555	65.7	99.0	28.93	(1.724)	1.204(0.839)	24.56	27.07	1.09[67.1]
Al	13	26.9815386(8)	0.48181	69.7	107.2	24.01	1.615	2.699	933.5	2792.	
Si	14	28.0855(3)	0.49848	70.2	108.4	21.82	1.664	2.329	1687.	3538.	3.95
Cl <sub>2</sub>	17	35.453(2)	0.47951	73.8	115.7	19.28	(1.630)	1.574(2.980)	171.6	239.1	[773.]
Ar	18	39.948(1)	0.45059	75.7	119.7	19.55	(1.519)	1.396(1.662)	83.81	87.26	1.23[281.]
Ti	22	47.867(1)	0.45961	78.8	126.2	16.16	1.477	4.540	1941.	3560.	
Fe	26	55.845(2)	0.46557	81.7	132.1	13.84	1.451	7.874	1811.	3134.	
Cu	29	63.546(3)	0.45636	84.2	137.3	12.86	1.403	8.960	1358.	2835.	
Ge	32	72.64(1)	0.44053	86.9	143.0	12.25	1.370	5.323	1211.	3106.	
Sn	50	118.710(7)	0.42119	98.2	166.7	8.82	1.263	7.310	505.1	2875.	
Xe	54	131.293(6)	0.41129	100.8	172.1	8.48	(1.255)	2.953(5.483)	161.4	165.1	1.39[701.]
W	74	183.84(1)	0.40252	110.4	191.9	6.76	1.145	19.300	3695.	5828.	
Pt	78	195.084(9)	0.39983	112.2	195.7	6.54	1.128	21.450	2042.	4098.	
Au	79	196.966569(4)	0.40108	112.5	196.3	6.46	1.134	19.320	1337.	3129.	
Pb	82	207.2(1)	0.39575	114.1	199.6	6.37	1.122	11.350	600.6	2022.	
U	92	[238.02891(3)]	0.38651	118.6	209.0	6.00	1.081	18.950	1408.	4404.	
Air (dry, 1 atm)			0.49919	61.3	90.1	36.62	(1.815)	(1.205)		78.80	
Shielding concrete			0.50274	65.1	97.5	26.57	1.711	2.300			
Borosilicate glass (Pyrex)			0.49707	64.6	96.5	28.17	1.696	2.230			
Lead glass			0.42101	95.9	158.0	7.87	1.255	6.220			
Standard rock			0.50000	66.8	101.3	26.54	1.688	2.650			

# Example

- Aluminum 1cm thickness with electron 50MeV
- Estimate the angle deviation at exit, ignore energy loss.
- Radiation length

$$X_0 = 24.01 \text{ [g/cm}^2\text{]} / 2.699 \text{ [g/cm}^3\text{]} = 8.9 \text{ [cm]}$$

$$x/X_0 = 0.11$$

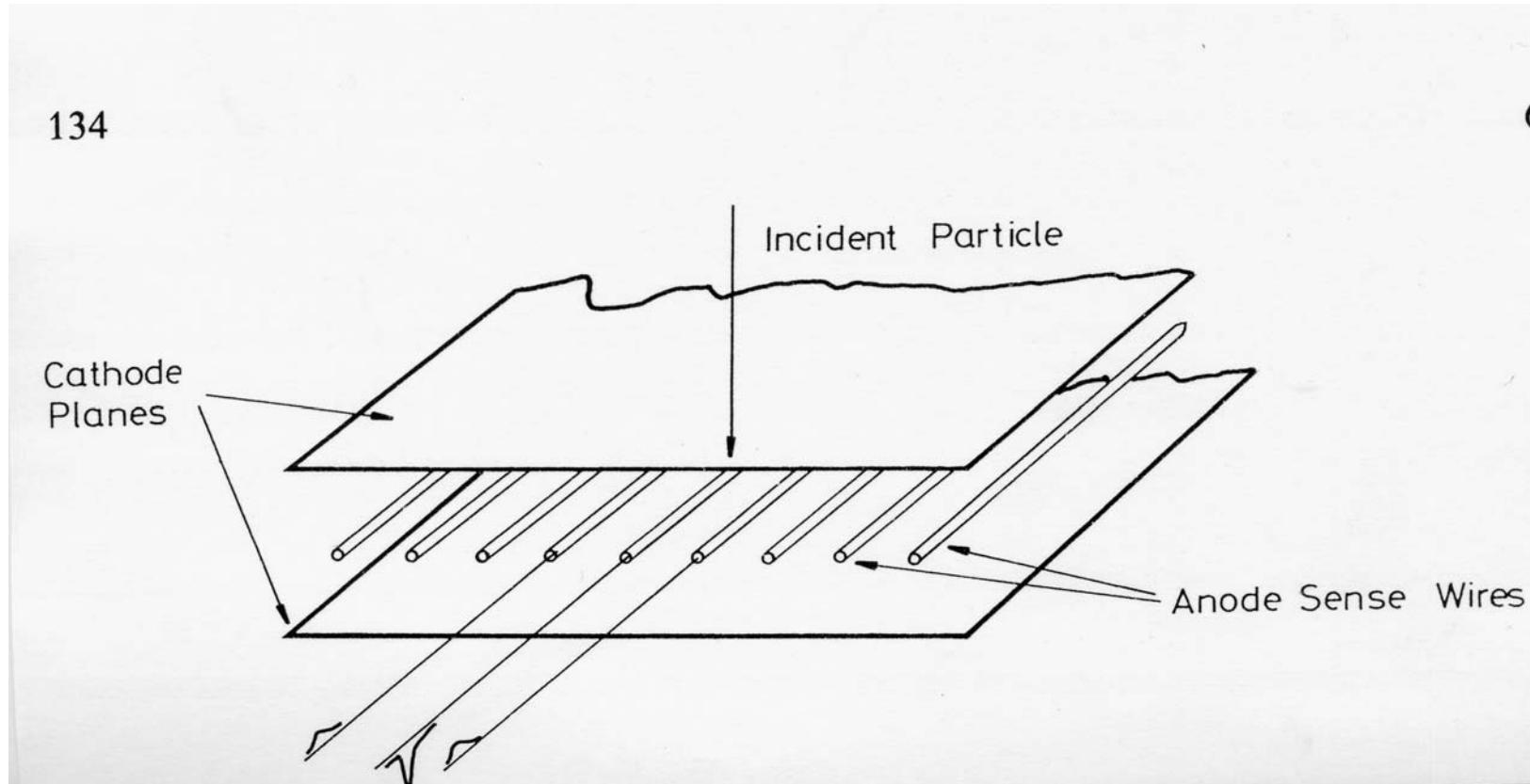
$$\theta_0 = \frac{13.6 \text{ MeV}}{\beta c p \text{ [MeV]}} z \sqrt{x/X_0} [1 + 0.038 \ln(x/X_0)]$$

$$= \frac{13.6}{50} * 1 * \sqrt{0.11} [1 + 0.038 \ln(0.11)]$$

$$= 0.082 \text{ [rad]} = 4.7 \text{ [deg]}$$

$$= \theta_{plane}^{rms}$$

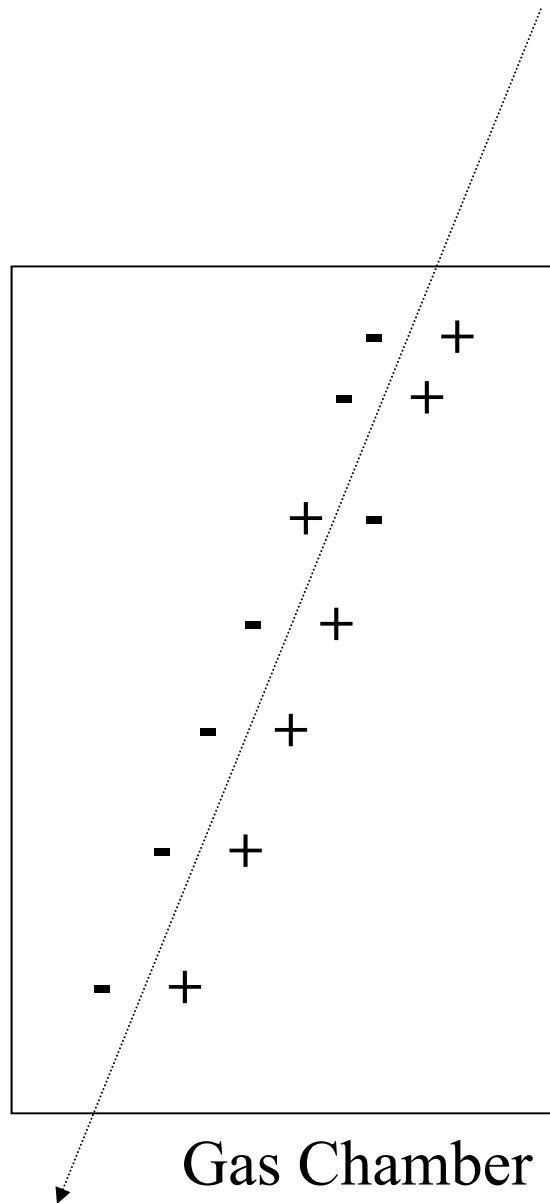
# Typical Wire Chamber



# Gas Chamber

1. Energetic particle passing through gas.
2. Gas molecules are ionized  $\rightarrow$  electrons and ions.
3. Electrons and ions are drifted to electro load with minus and plus voltage
4. Avalanche near by wire
5. Getting electrical signal

# Ionization



- Ionization happens along charged particle track
- #Electron-Ion pair/unit length =  $dE/dX$  / Pair creation energy
- Pair creation Energy
  - H<sub>2</sub> 37eV
  - Ar 26eV

# Ionization

**Table 6.1.** Excitation and ionization characteristics of various gases

	Excitation potential [eV]	Ionization potential [eV]	Mean energy for ion-electron pair creation [eV]
H <sub>2</sub>	10.8	15.4	37
He	19.8	24.6	41
N <sub>2</sub>	8.1	15.5	35
O <sub>2</sub>	7.9	12.2	31
Ne	16.6	21.6	36
Ar	11.6	15.8	26
Kr	10.0	14.0	24
Xe	8.4	12.1	22
CO <sub>2</sub>	10.0	13.7	33
CH <sub>4</sub>		13.1	28
C <sub>4</sub> H <sub>10</sub>		10.8	23

Ar:  $dE/dX = 1.519 \text{ MeV cm}^2/\text{g}$       electron-ion = 97 /cm

density = 1.662 g/L

pair creation energy = 26 eV

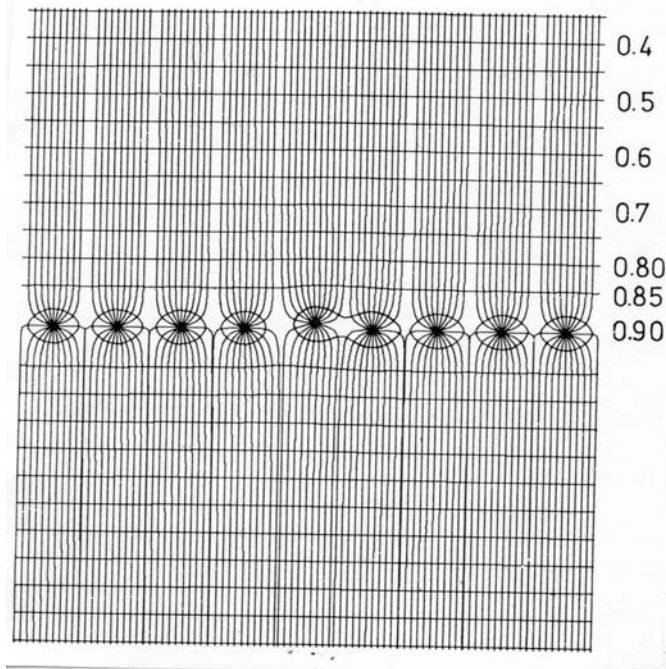
#electron-ion =  $1.519 \text{ MeV cm}^2/\text{g} * 106 \text{ eV/MeV} * 1.662 \text{ g/l} * 1000 \text{ cm}^3/\text{l} / 26 \text{ eV}$

= 97 /cm

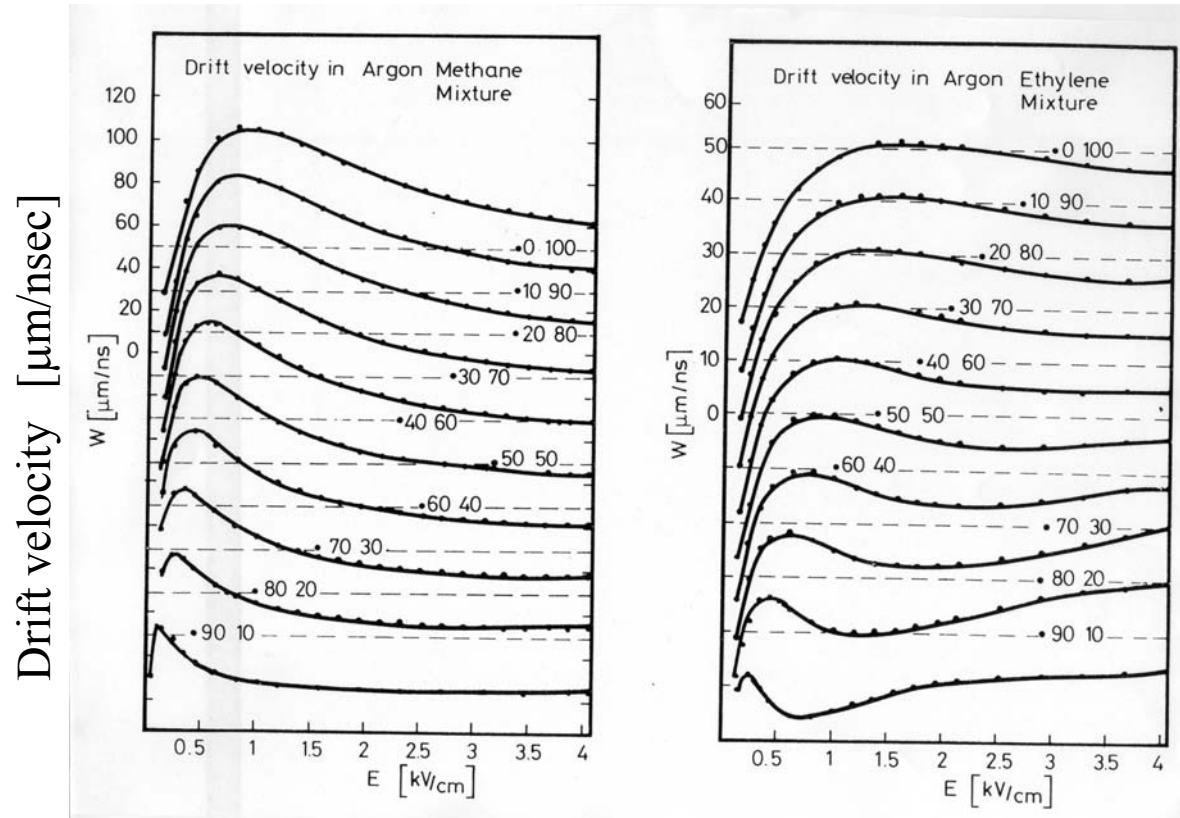
# Drift

drift velocity =  $50 \sim 100 \mu\text{m/nsec}$

=  $5\text{mm} \sim 10\text{mm} / 100 \text{ nsec}$



Electric field  
and potential



Electric field [KV/cm]

# Avalanche

$$E = \frac{1}{r} \frac{V_0}{\ln(b/a)}$$

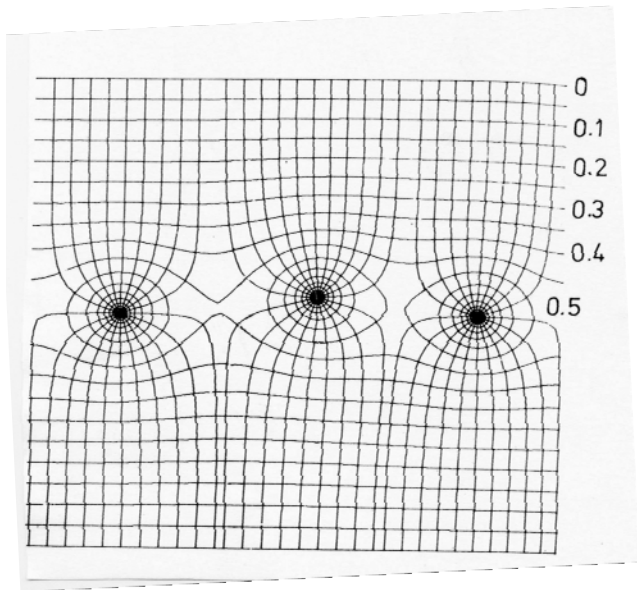
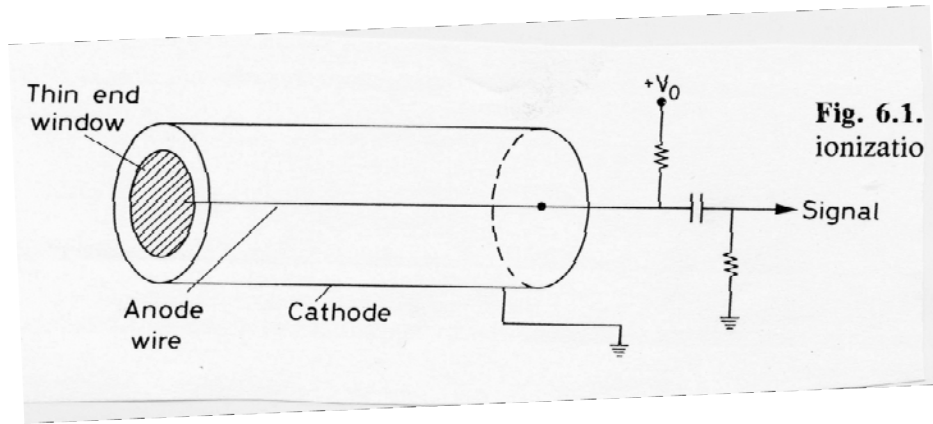
E: electric field

r: distance from wire

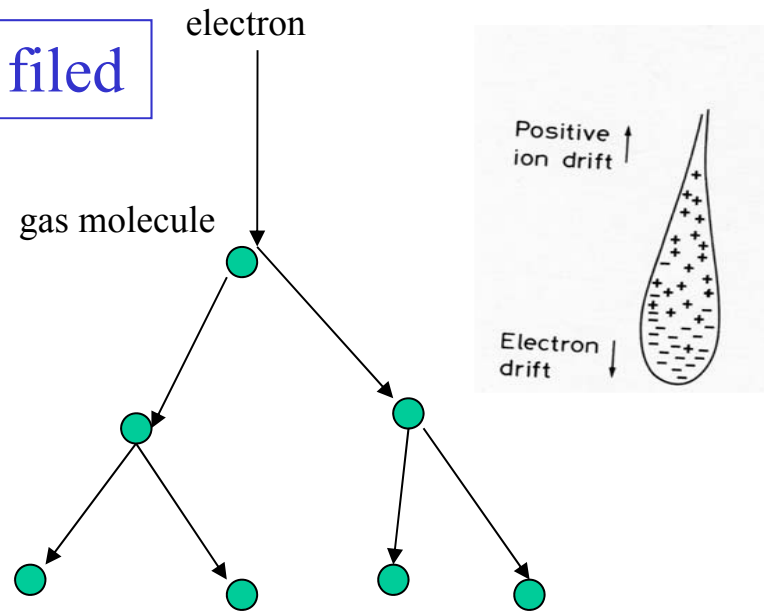
V<sub>0</sub>: bias voltage

a: radius of cylinder

b: wire radius



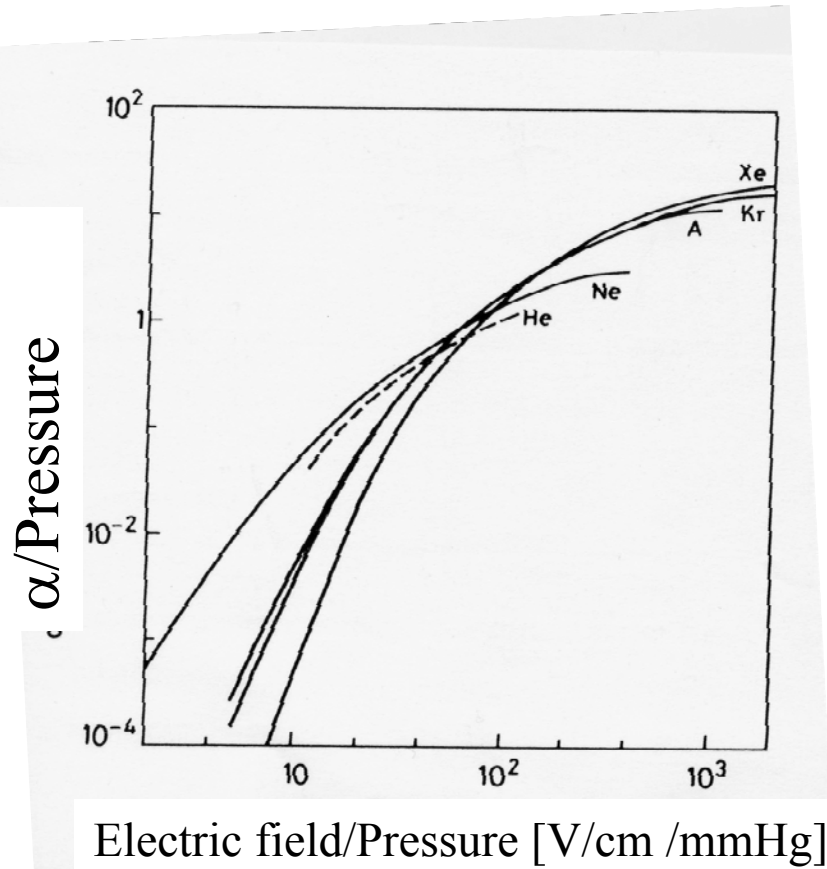
Electric field



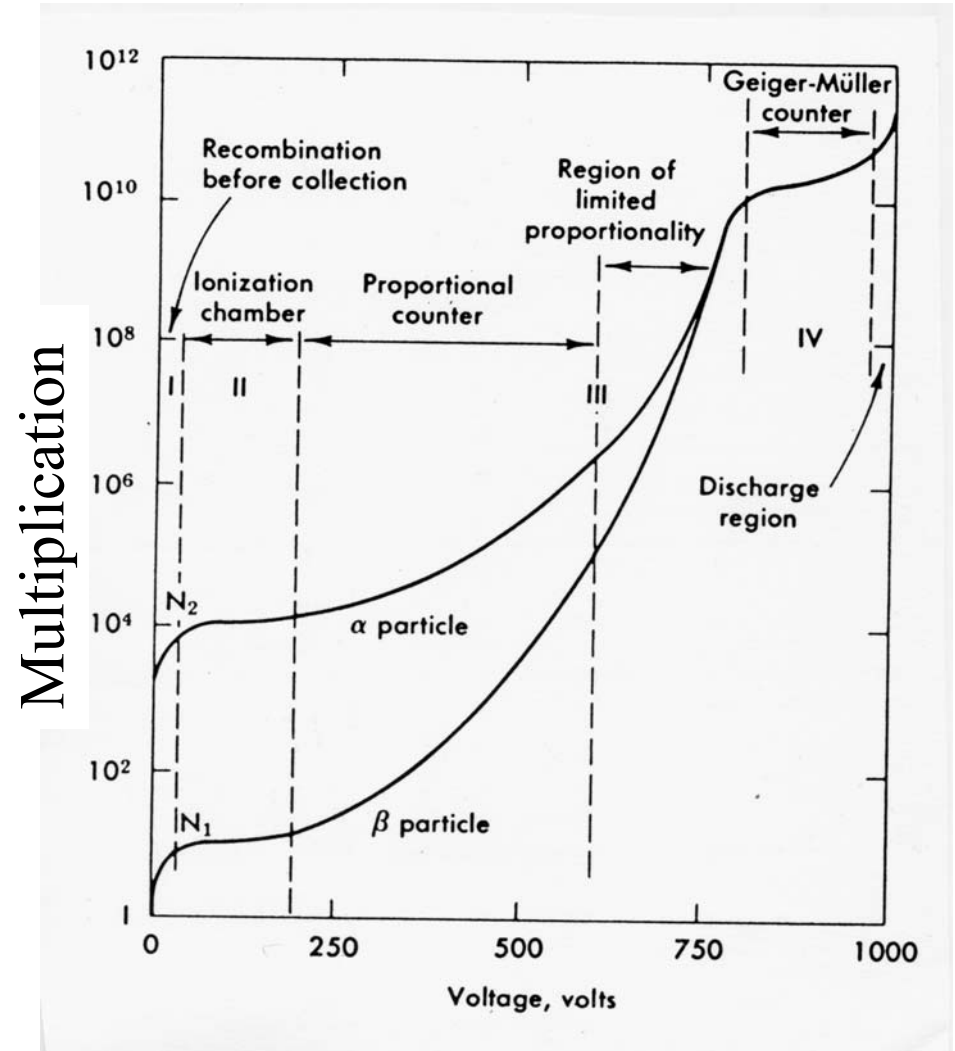
Gas multiplication



# Gas Gain

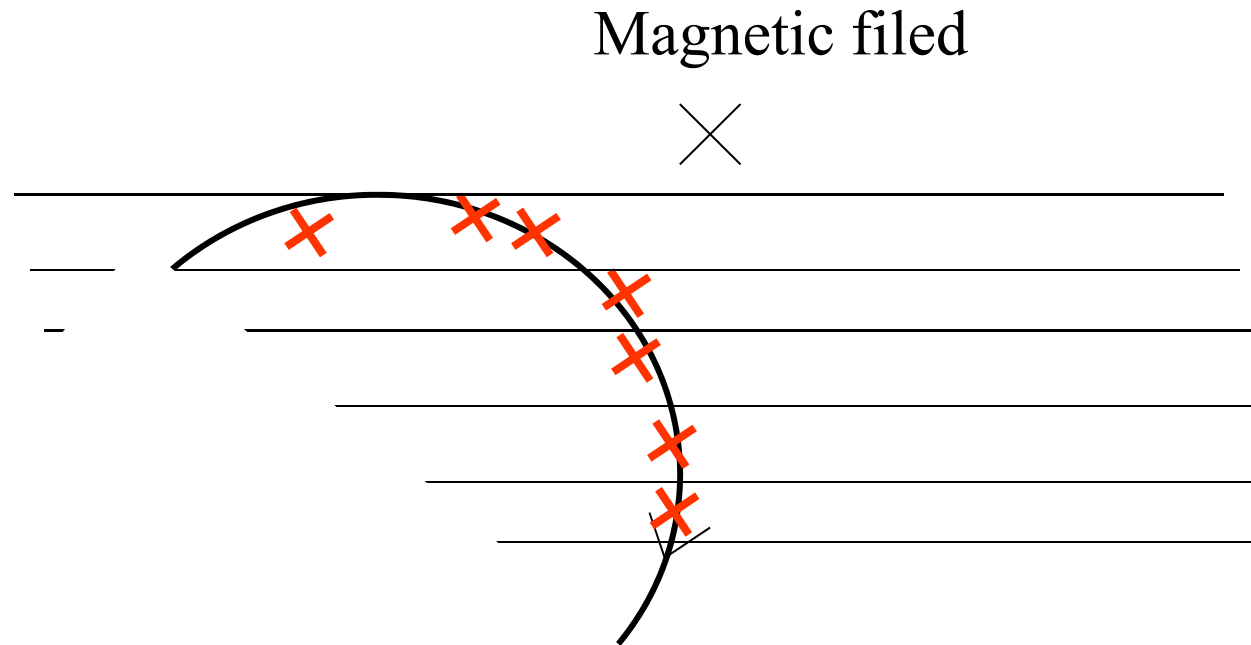


$$\text{Multiplication} = \exp \left[ \int_{r_1}^{r_2} \alpha(r) dr \right]$$



Voltage [V]

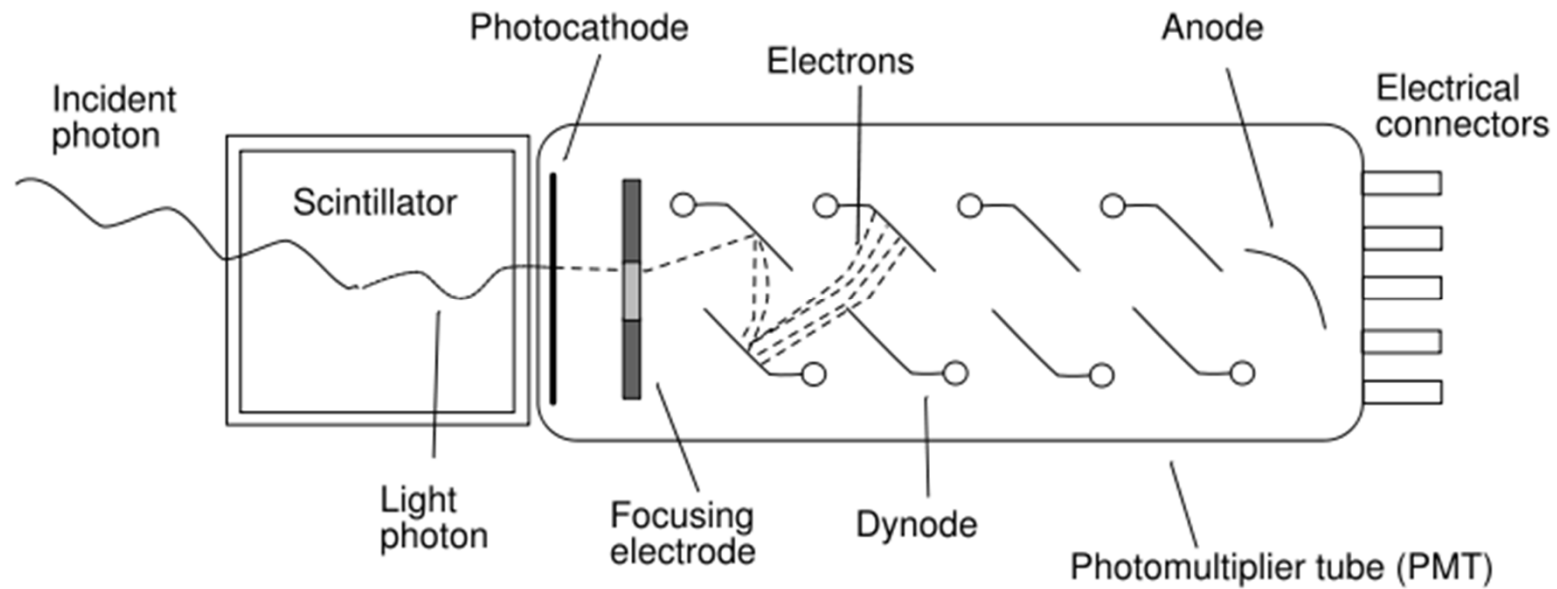
# Measurement of the momentum



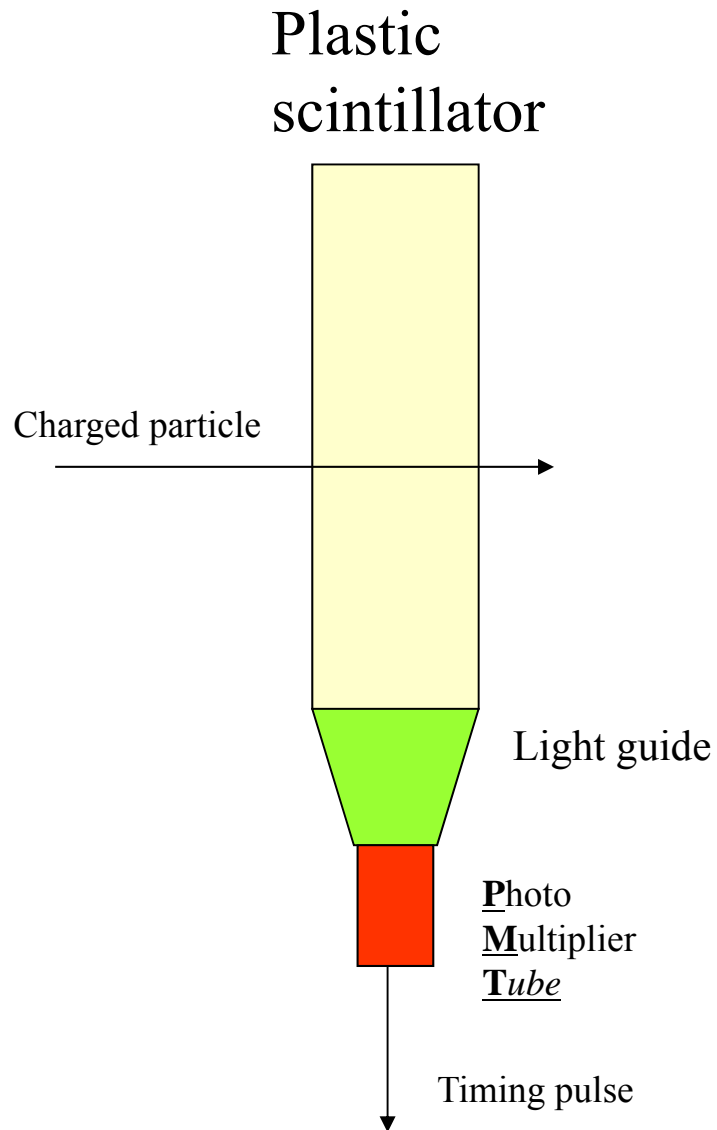
$$P [\text{MeV}/c] = 3 * r[\text{cm}] * B[\text{T}]$$

r: curvature radius, B: magnetic field

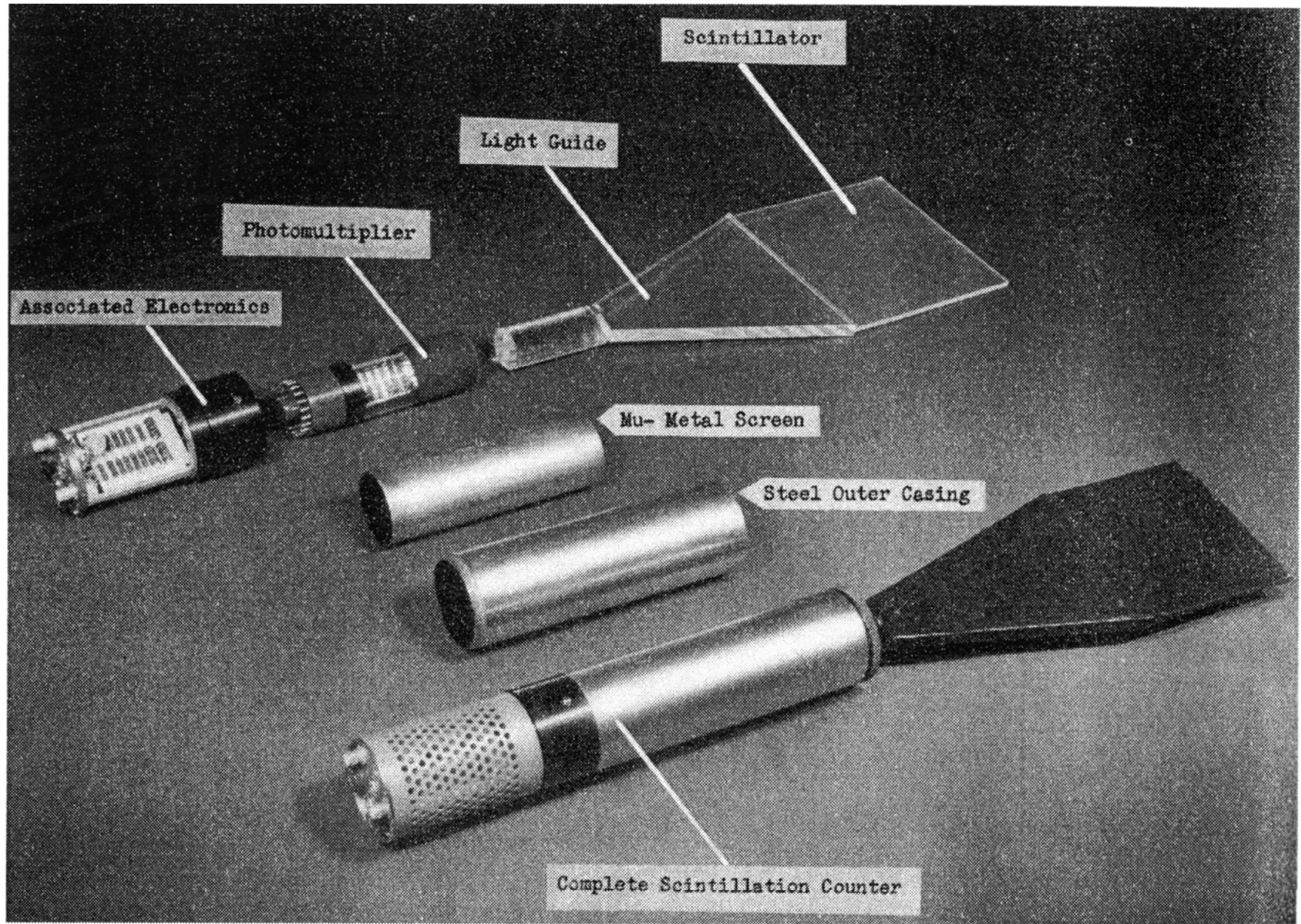
# Photo Multiplier Tube



# Precise Time measurement detector



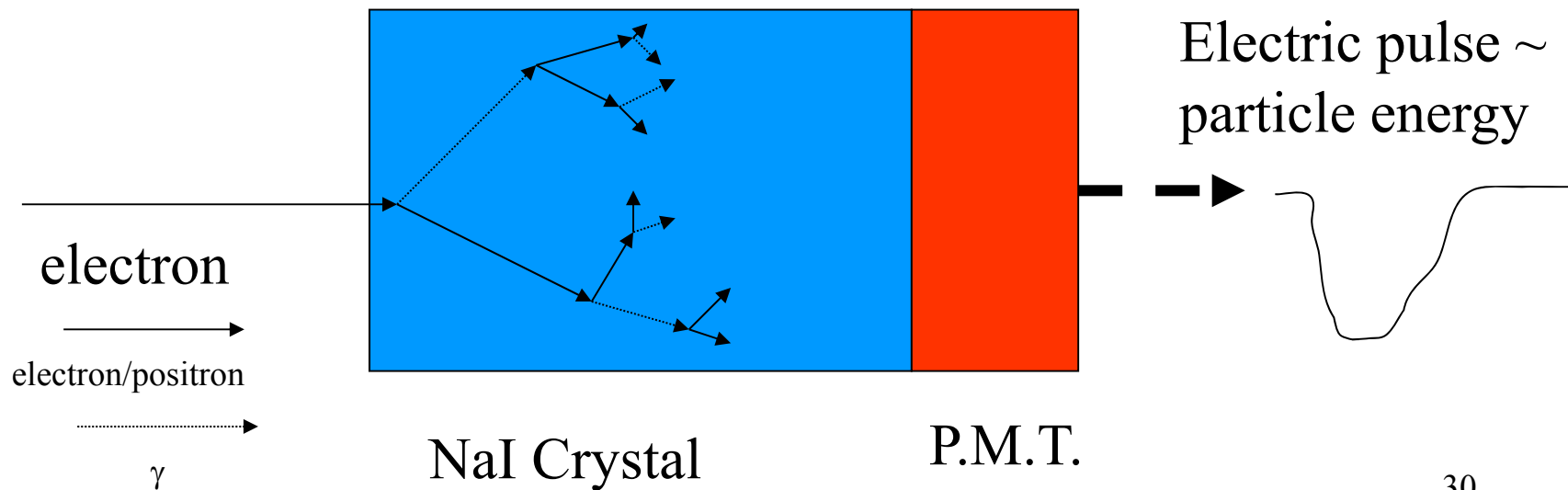
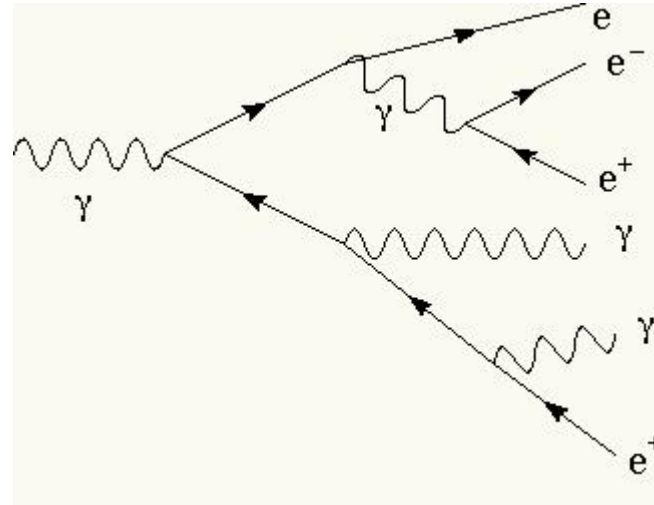
1. Charged particle is penetrating
2. Lower energy electron is excited to upper state
3. Upper state electron drops into lower state and emits a photon
4. Propagate to P.M.T.
5. P.M.T. generates electric pulse.



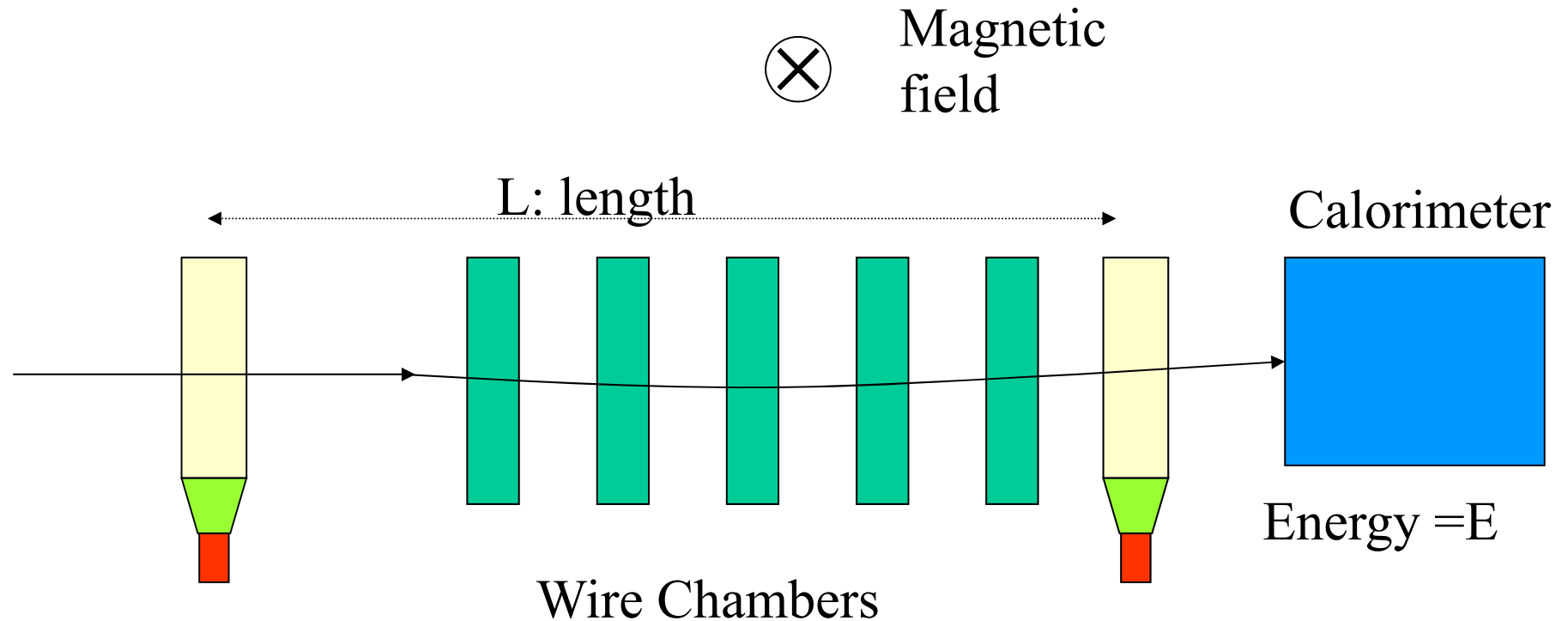
# Calorimeter

Mass of electron is 511KeV

$\gamma \rightarrow e^+ + e^-$  for  $E_\gamma > 1.02\text{MeV}$



# Typical Detector System



Scintillator

Timing =  $t_1$

$$P [\text{MeV}/c] = 3 * r[\text{cm}] * B[\text{T}]$$

$r$ : curvature radius,  $B$ : magnetic field

Scintillator

Timing =  $t_2$

$$\beta = v/c = L/(t_2 - t_1)/c$$

$$E = P/\beta$$

$$m^2 = E^2 + P^2 \longrightarrow \text{Determine 4-momentum}$$

# Other Major Detectors (include past)

Detector Type	Accuracy (rms)	Resolution Time	Dead Time
Bubble chamber	10–150 $\mu\text{m}$	1 ms	50 ms <sup>a</sup>
Streamer chamber	300 $\mu\text{m}$	2 $\mu\text{s}$	100 ms
Proportional chamber	50–300 $\mu\text{m}^{b,c,d}$	2 ns	200 ns
Drift chamber	50–300 $\mu\text{m}$	2 ns <sup>e</sup>	100 ns
Scintillator	—	100 ps/n <sup>f</sup>	10 ns
Emulsion	1 $\mu\text{m}$	—	—
Liquid Argon Drift [Ref. 6]	$\sim$ 175–450 $\mu\text{m}$	$\sim$ 200 ns	$\sim$ 2 $\mu\text{s}$
Gas Micro Strip [Ref. 7]	30–40 $\mu\text{m}$	< 10 ns	—
Resistive Plate chamber [Ref. 8]	$\lesssim$ 10 $\mu\text{m}$	1–2 ns	—
Silicon strip	pitch/(3 to 7) <sup>g</sup>	$h$	$h$
Silicon pixel	2 $\mu\text{m}^i$	$h$	$h$

Position  
sensitivity

Timing  
sensitivity

Detector dead  
time after a hit



# Summary

- We need to have detector to investigate nature since you can not feel particles.
- You have to build and/or be familiar with detector for your own experiments.

# PHENIX Where is VTX

